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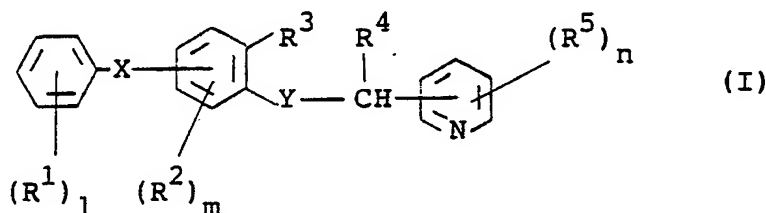
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Siebertstrasse 4 P.O. Box 86 07 67
W-8000 München 86(DE)(54) **Pyridine derivatives, their production processes and their compositions for the control of insect pests.**

(57) A pyridine derivative of the formula (I)



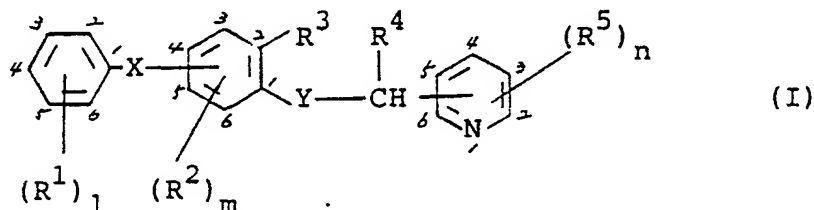
wherein R¹ is, the same or different, each a hydrogen atom, a halogen atom, a C₁-C₄ alkyl group, a C₁-C₄ haloalkyl group, a C₁-C₃ alkoxy group, a C₁-C₃ haloalkoxy group, a cyano group or a nitro group; R² is, the same or different, each a hydrogen atom, a halogen atom or a methyl group; R³ is a halogen atom or a C₁-C₃ alkyl group; R₄ is a hydrogen atom or a C₁-C₃ alkyl group; R⁵ is, the same or different, each a halogen atom, a C₁-C₃ alkyl group, a C₁-C₃ haloalkyl group, a C₁-C₃ alkoxy group or a C₁-C₃ haloalkoxy group; X is an oxygen atom, a sulfur atom, a carbonyl group, a sulfinyl group, a sulfonyl group, an imino group or a methylene group; Y is an oxygen atom or a sulfur atom; l is an integer of 1 to 5; m is an integer of 1 to 3; and n is an integer of 1 to 4, which is useful for the control of insect pests.

The present invention relates to pyridine derivatives, their production processes and their compositions for the control of insect pests.

It is described in German Offenlegungsschrift 2,516,331 that certain pyridine derivatives are useful as insecticides and acaricides. But, their insecticidal and acaricidal activities are still not satisfactory.

Thus, it is the object of the present invention to provide pyridine compounds producing a satisfactory controlling effect on insect pests.

This object has been achieved by the surprising finding that pyridine derivatives of the following formula (I) exhibit a remarkable juvenile hormone-like activity and can control significantly the growth of insect pests:



wherein R¹ is, the same or different, each a hydrogen atom, a halogen atom, a C₁-C₄ alkyl group, a C₁-C₄ haloalkyl group, a C₁-C₃ alkoxy group, a C₁-C₃ haloalkoxy group, a cyano group or a nitro group; R² is, the same or different, each a hydrogen atom, a halogen atom or a methyl group; R³ is a halogen atom or a C₁-C₃ alkyl group; R⁴ is a hydrogen atom or a C₁-C₃ alkyl group; R⁵ is, the same or different, each a halogen atom, a C₁-C₃ alkyl group, a C₁-C₃ haloalkyl group, a C₁-C₃ alkoxy group or a C₁-C₃ haloalkoxy group; X is an oxygen atom (-O-), a sulfur atom (-S-), a carbonyl group (-CO-), a sulfinyl group (-SO-), a sulfonyl group (-SO₂-), an imino group (-NH-) or a methylene group (-CH₂-); Y is an oxygen atom (-O-) or a sulfur atom (-S-); l is an integer of 1 to 5; m is an integer of 1 to 3; and n is an integer of 1 to 4.

The pyridine derivatives (I) of the invention have an excellent juvenile hormone-like activity against insect pests. They exhibit various actions such as metamorphosis inhibition, embryogenesis inhibition and sterilization and are thus efficacious as growth regulators, chemosterilants, ovicides or reproduction inhibitory agents on various insect pests such as agricultural, forestal, hygienic and stored grain insect pests. They are also efficacious against insect pests having an increased resistance to commercial insecticides.

In the formula (I) which represents the pyridine derivatives of the invention, examples of the halogen atom are fluorine, chlorine, bromine, iodine, etc.

Examples of the C₁-C₃ alkyl group are methyl, ethyl, n-propyl, isopropyl, etc., while examples of the C₁-C₄ alkyl group are methyl, ethyl, n-propyl, isopropyl, n-butyl, sec-butyl, isobutyl, tert-butyl, etc. Examples of the C₁-C₃ haloalkyl group include trifluoromethyl, difluoromethyl, 2-fluoroethyl, 1-fluoroethyl, 1,2-difluoroethyl, 2,2,2-trifluoroethyl, 2-chloroethyl, 3-fluoro-n-propyl, 2-fluoro-n-propyl, 1-fluoro-n-propyl, 3-chloro-n-propyl, 3-bromo-n-propyl, etc., while examples of the C₁-C₄ haloalkyl group include trifluoromethyl, difluoromethyl, 2-fluoroethyl, 1-fluoroethyl, 1,2-difluoroethyl, 2,2,2-trifluoroethyl, 2-chloroethyl, 3-fluoro-n-propyl, 2-fluoro-n-propyl, 1-fluoro-n-propyl, 3-chloro-n-propyl, 3-bromo-n-propyl, 4-fluoro-n-butyl, 4-chloro-n-butyl, etc. Examples of the C₁-C₃ alkoxy group include methoxy, ethoxy, n-propoxy, isopropoxy, etc. Examples of the C₁-C₃ haloalkoxy group include trifluoromethoxy, difluoromethoxy, bromodifluoromethoxy, 2-fluoroethoxy, 2,2,2-trifluoroethoxy, 1-fluoroethoxy, 1,2-difluoroethoxy, 3-fluoro-n-propoxy, 1-fluoro-n-propoxy, 2-fluoro-n-propoxy, 2-chloroethoxy, 3-chloro-n-propoxy, 3-bromo-n-propoxy, 1,1,2,2-tetrafluoroethoxy, 1,1-difluoromethoxy, etc.

Among the pyridine derivatives (I), preferred are those wherein R¹ is, the same or different, each a hydrogen atom, a fluorine atom, a chlorine atom, a bromine atom, a methyl group or a trifluoromethyl group. More preferred are those wherein R¹ is, the same or different, each a hydrogen atom, a fluorine atom or a chlorine atom, R² is a hydrogen atom or a chlorine atom, R³ is a fluorine atom, a chlorine atom, a bromine atom or a methyl group, X is an oxygen atom or a methylene group, Y is an oxygen atom, l is an integer of 1 or 2, m is an integer of 1 and n is an integer of 1 or 2. Further more preferred are those wherein (R¹)_l is 2,4-F₂, 3,5-F₂, 3,4-Cl₂, 3-F or 3-Cl, R² is a hydrogen atom, R₃ is a chlorine atom, R⁴ is a hydrogen atom, R⁵ is a chlorine atom or a methyl group, X is an oxygen atom or a methylene group, Y is an oxygen atom and m and n are each an integer of 1. When X is a methylene group, R¹ may be additionally a hydrogen atom. Most preferred are those wherein (R¹)_l is 2,4-F₂, 3,5-F₂ or 3-Cl, R² is a hydrogen atom, R³ is a

chlorine atom, R^4 is a hydrogen atom, R^5 is a chlorine atom or a methyl group, X is an oxygen atom or a methylene group, Y is an oxygen atom and m and n are each an integer of 1. When X is a methylene group, (R^1)₁ may be additionally 3-F.

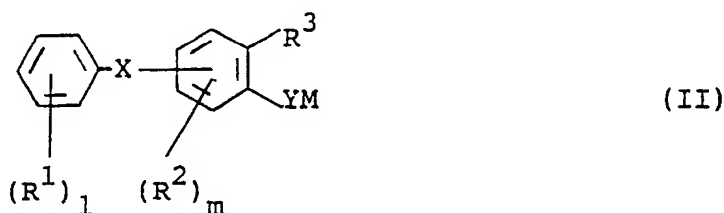
The optionally substituted phenyl-X- group may be present at any of the 3- to 6-positions on the benzene ring but the 4- or 5-position is preferred.

Also, the R^4 -bearing methylene group may be attached to the substituted pyridyl group at any position of the latter but the 4- or 5-position, particularly the 5-position, is preferred. Further, the substituted pyridyl group may bear one or more substituents at any position, preferably at least at the 2-position.

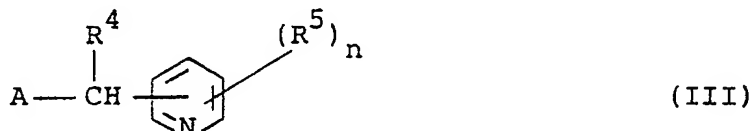
The pyridine derivatives (I) of the invention can be produced by various processes, among which typical examples are shown below.

Process A:-

The pyridine derivative (I) is produced by reacting a phenol compound of the formula (II)



wherein M is an alkali metal atom or a hydrogen atom, R^1 , R^2 , R^3 , X, Y, I and m are each as defined above with a halide of the formula: (III)



wherein A is a halogen atom and R^4 , R^5 and n are each as defined above.

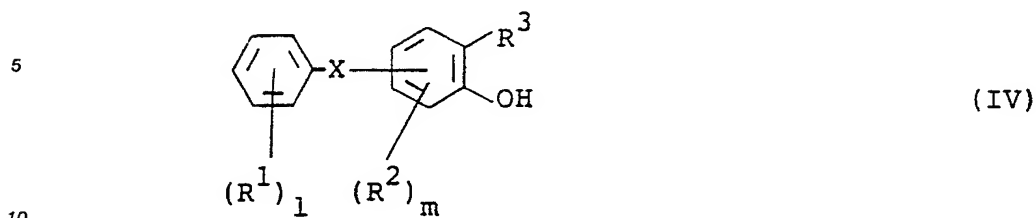
The reaction may be carried out usually in an inert solvent in the presence of a base at a temperature of from about -20°C to the boiling point of the inert solvent, preferably from about -5°C to the boiling point of the inert solvent. When the phenol compound (II) is employed in a metal salt form, the base is not necessarily required to use.

The molar proportion of the phenol compound (II) and the halide (III) to be used for the reaction is not limitative but is preferred to be nearly equal. Examples of the inert solvent are lower alcohols (e.g. methanol, ethanol, propanol, isopropanol, tert-butanol), ketones (e.g. acetone, methyl ethyl ketone), hydrocarbons (e.g. hexane, benzene, toluene), ethers (e.g. diethyl ether, diisopropyl ether, tetrahydrofuran, dioxane), acid amides (e.g. N,N-dimethylformamide, hexamethylphosphoric triamide), halogenated hydrocarbons (e.g. dichloromethane, chloroform, 1,2-dichloroethane, chlorobenzene), nitriles (e.g. acetonitrile), nitro compounds (e.g. nitromethane), dimethyl sulfoxide, water and mixtures thereof. Examples of the base are alkali metal hydroxides (e.g. sodium hydroxide, potassium hydroxide), alkali metal carbonates (e.g. potassium carbonate), alkali metals (e.g. metallic sodium), alkali metal hydrides (e.g. sodium hydride), organic bases (e.g. sodium methoxide, sodium ethoxide, triethylamine, pyridine), etc. When necessary or desired, an ammonium salt such as triethylbenzylammonium chloride may be added to the reaction system as a catalyst.

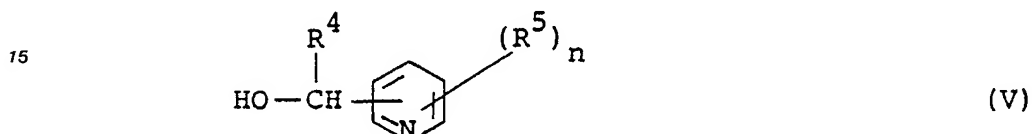
After completion of the reaction, post-treatment may follow in a per se conventional manner such as extraction with an organic solvent and concentration. When necessary or desired, the product may further be purified by chromatography, distillation, recrystallization, etc.

Process B:-

The pyridine derivative (I) is produced by reacting a phenol compound of the formula (IV)



wherein R^1 , R^2 , R^3 , X, l and m are each as defined above with an alcohol of the formula (V)



wherein R^4 , R^5 and n are each as defined above. This process is advantageously applicable when R^4 is other than a hydrogen atom.

The reaction may be carried out normally in an inert solvent in the presence of a dehydrating catalyst or agent at a temperature of from about -20 to 200 °C or the boiling point of the inert solvent.

The molar proportion of the phenol compound (IV) and the alcohol (V) is not limitative but is preferred to be nearly equal.

Examples of the dehydrating catalyst are inorganic acids (e.g. hydrochloric acid, sulfuric acid), aromatic sulfonic acids, sulfonic acid halides, etc. Examples of the dehydrating agent include dicyclohexylcarbodiimide, di-isopropylazodicarboxylate, diethylazodicarboxylate, etc. Examples of the inert solvent are hydrocarbons (e.g. benzene, toluene, xylene), ethers (e.g. diethyl ether, diisopropyl ether, tetrahydrofuran, dioxane), halogenated hydrocarbons (e.g. carbon tetrachloride, dichloromethane, 1,2-dichloroethane, chlorobenzene, dichlorobenzene), etc.

After completion of the reaction, post-treatment may follow in a per se conventional manner such as extraction with an organic solvent and concentration. When necessary or desired, the product may further be purified by chromatography, distillation, recrystallization, etc.

The pyridine derivatives (I) of the invention have some asymmetric carbon atoms and can form optical isomers. Those optical isomers and their mixtures fall within the scope of the invention.

Among the starting compounds in the above processes, the phenol compound (II) wherein M is a hydrogen atom can be produced, for instance, as shown in the following scheme:

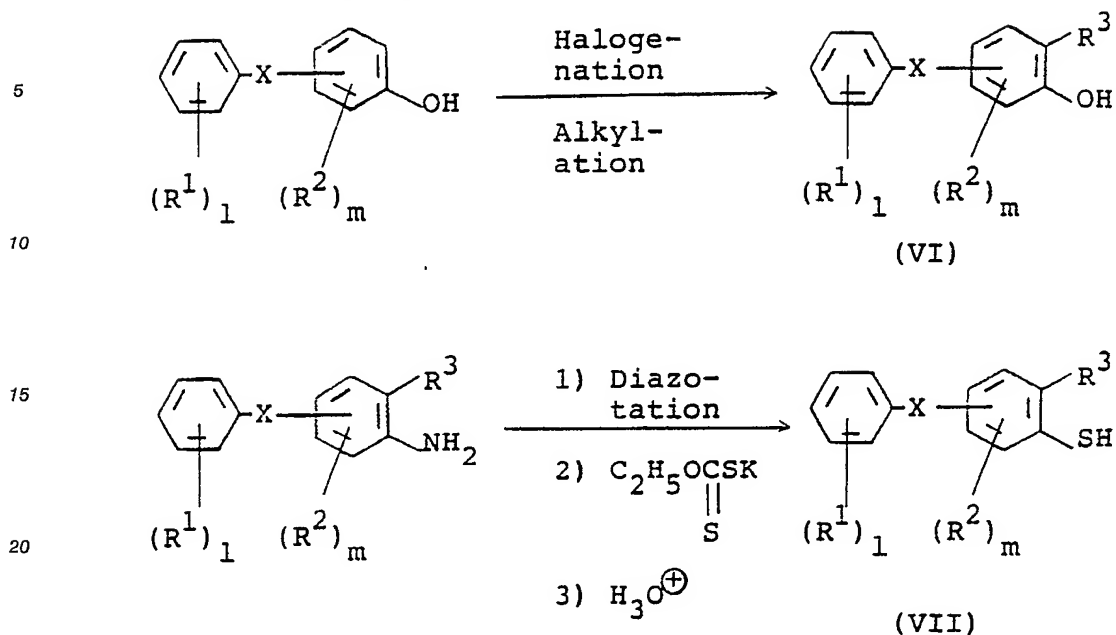
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Reaction Scheme 1



wherein R^1 , R^2 , R^3 , X , l and m are each as defined above.

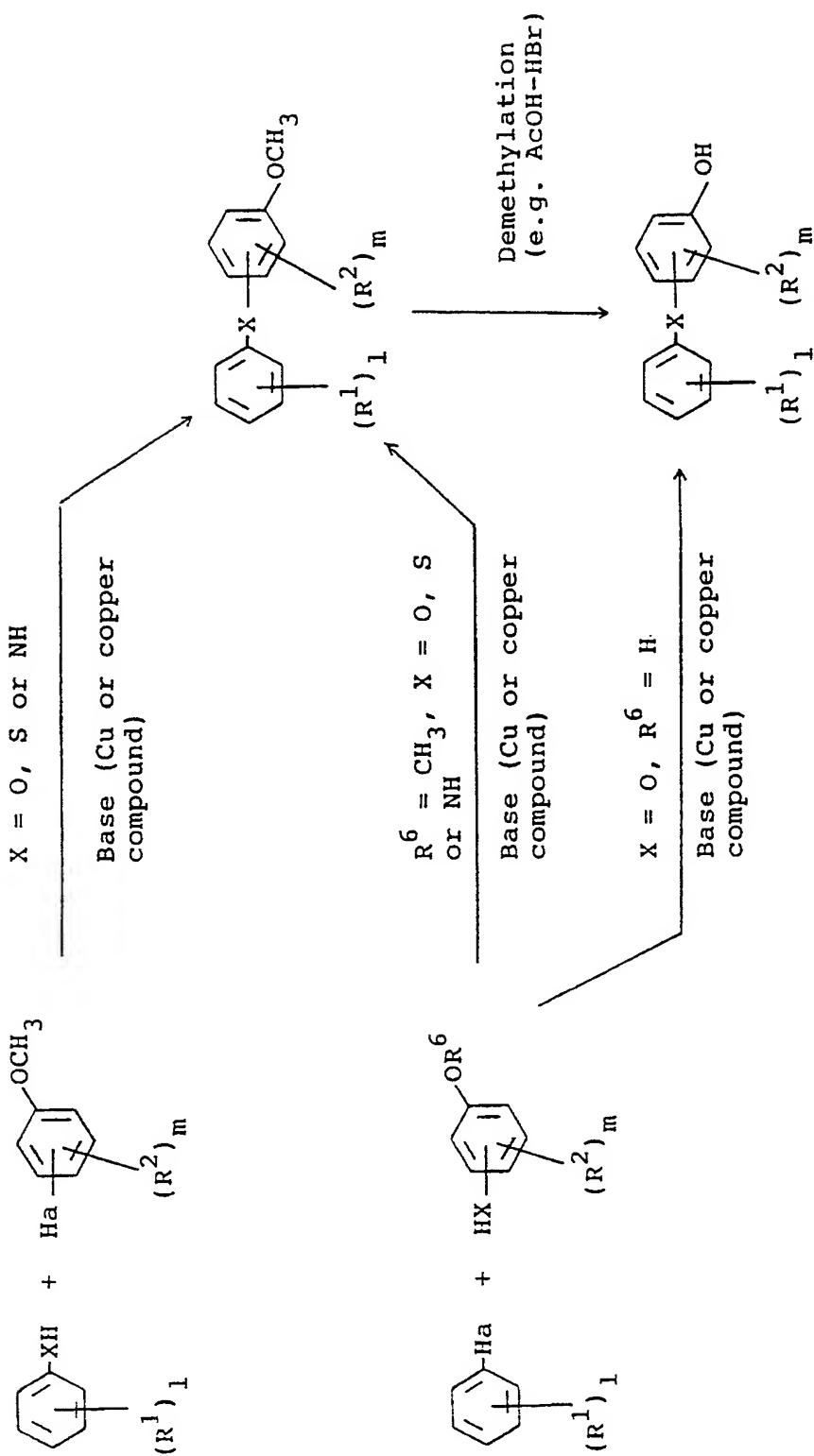
In the above scheme, each reaction may be performed by a per se conventional procedure. For instance, the phenol compound (VI) wherein R^3 is a chlorine atom may be prepared by reacting the corresponding non-chlorinated compound (i.e. the compound of the formula (VI) wherein R^3 is a hydrogen atom) with a chlorinating agent, preferably in an inert solvent. The molar proportion of the starting non-chlorinated compound and the chlorinating agent is not limitative, but it is ordinary to use the chlorinating agent in an amount equivalent to the non-chlorinating compound or somewhat in excess. Examples of the chlorinating agent are chlorine, tert-butylhydrochlorous acid, sulfonyl chloride, etc. If necessary and desired, the reaction can be carried out in the presence of a solvent. Examples of the inert solvent are dichloromethane, 1,2-dichloroethane, carbon tetrachloride, benzene, acetic acid, etc. The chlorinating agent itself may be available as a reaction medium when it is liquid. The reaction temperature is usually from about $-80^\circ C$ to the refluxing temperature of the reaction system, preferably from about $-20^\circ C$ to the refluxing temperature of the reaction system.

After completion of the reaction, post-treatment may follow in a per se conventional manner such as extraction with an organic solvent and concentration. When necessary or desired, the product may further be purified by chromatography, distillation, recrystallization, etc.

Said starting non-chlorinated compound, i.e. the compound of the formula (VI) wherein R^3 is a hydrogen atom, can be produced according to the reaction scheme as shown in Reaction Scheme 2.

Reaction Scheme 2

X = O, S or NH:-



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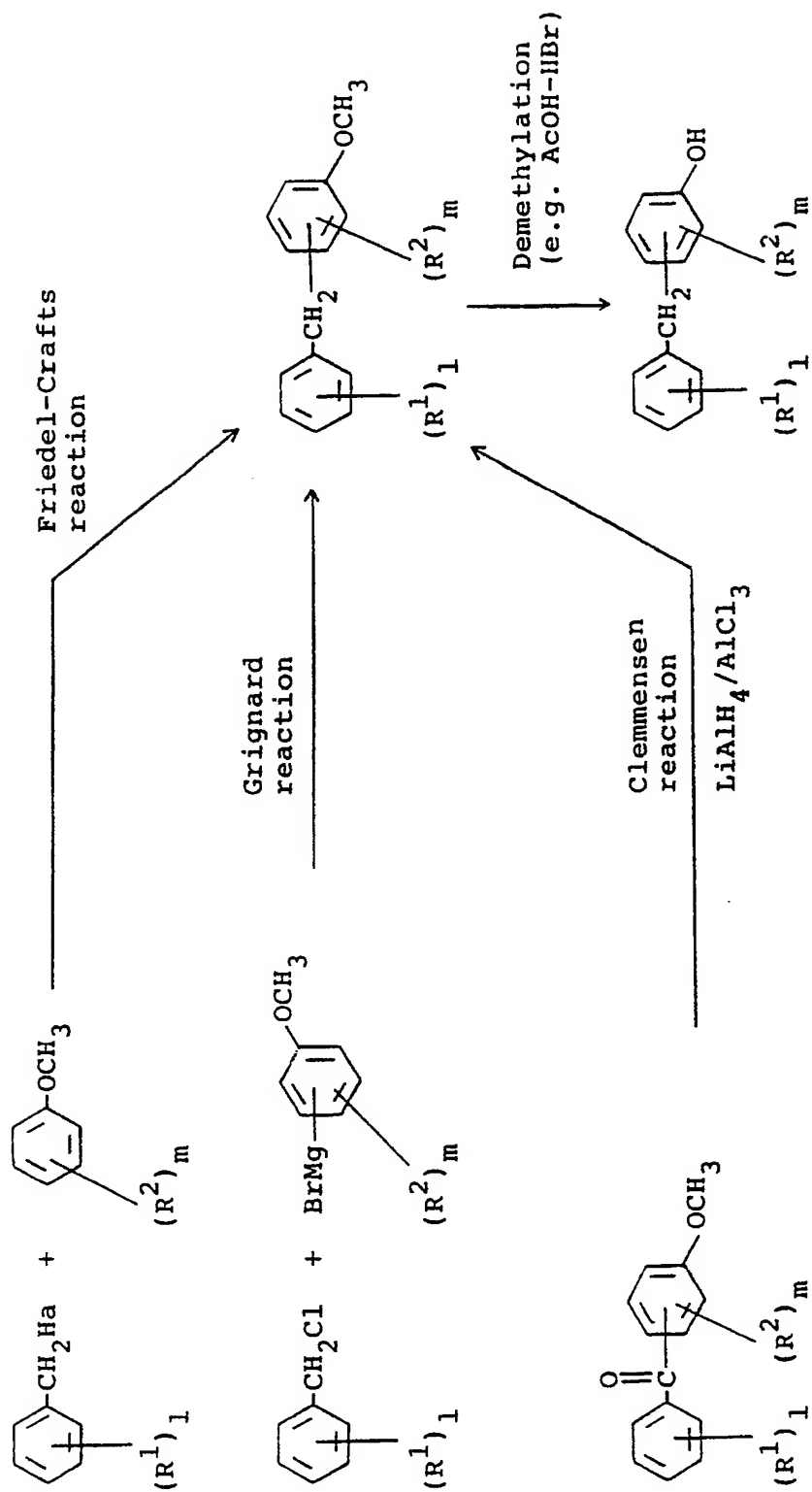
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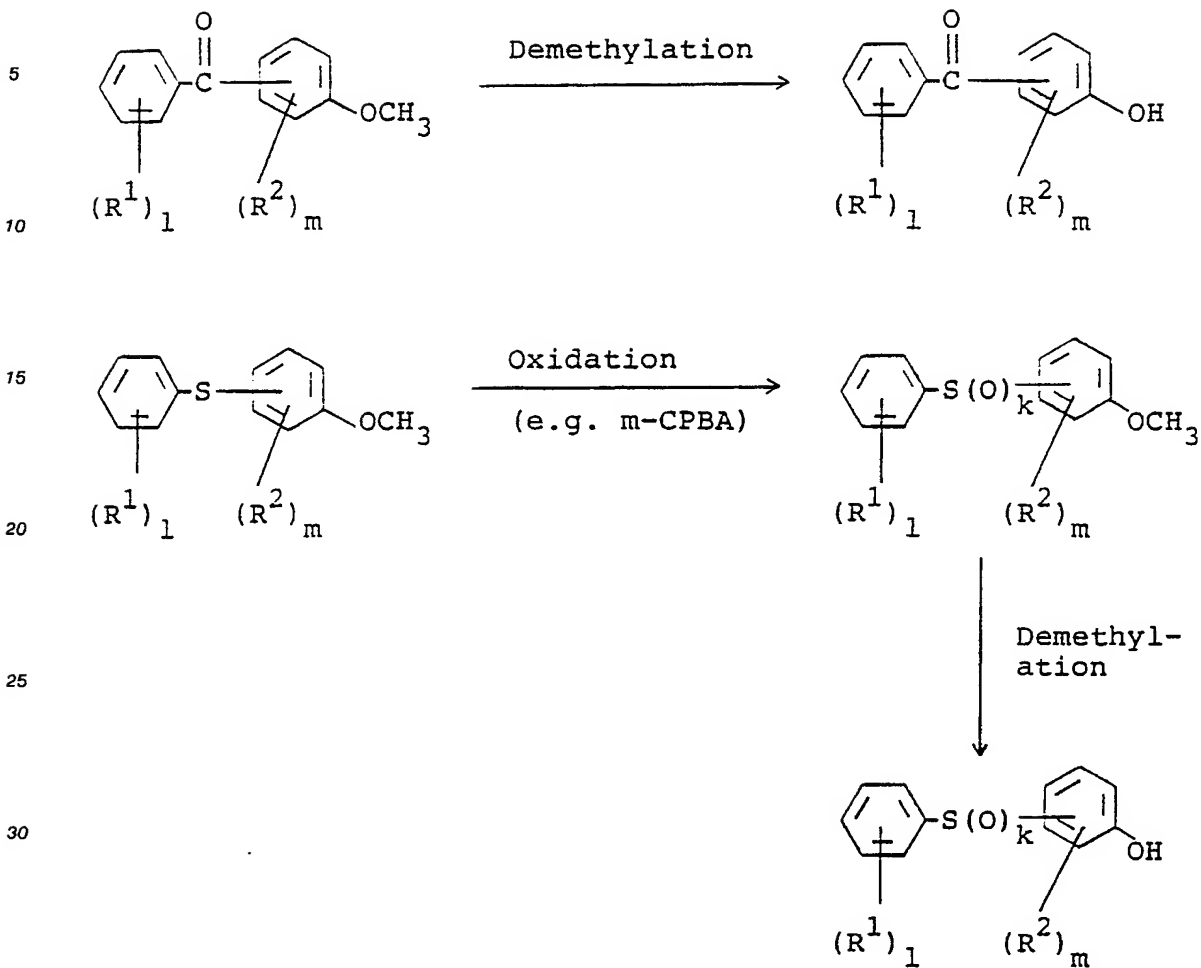
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 $X = CH_2:-$


X = CO, SO or SO₂:-



wherein R¹ R², l and m are each as defined above, Ha is a halogen atom and k is an integer of 1 or 2.
Examples of the pyridine derivatives (I) of the present invention are shown in Table 1.

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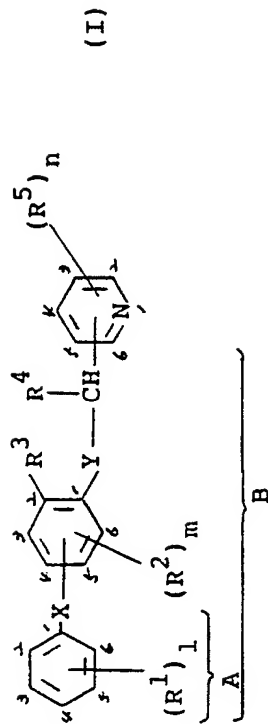
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Table 1



$(R^1)_1$	\underline{X}	$\underline{R^2}_m$	$\underline{R^3}$	\underline{Y}	$\underline{R^4}$	$\underline{(R^5)_n}$	\underline{A}	\underline{B}
H	O	H	Cl	O	H	2-Cl	4	5
H	O	H	F	O	H	2-Cl	4	5
H	O	H	Br	O	H	2-Cl	4	5
H	O	H	CH ₃	O	H	2-Cl	4	5
H	O	H	C ₂ H ₅	O	H	2-Cl	4	5
H	O	H	(n)C ₃ H ₇	O	H	2-Cl	4	5
3-F	O	H	Cl	O	H	2-Cl	4	5
3-F	O	H	F	O	H	2-Cl	4	5
3-F	O	H	Br	O	H	2-Cl	4	5
3-F	O	H	CH ₃	O	H	2-Cl	4	5
4-F	O	H	Cl	O	H	2-Cl	4	5

5	(continued)	$\frac{A}{B}$	$\frac{(R^5)}{(R^5)}_n$	$\frac{R^4}{R^4}$	$\frac{Y}{Y}$	$\frac{R^3}{R^3}$	$\frac{R^2}{R^2}_m$	$\frac{X}{X}$	$\frac{(R^1)}{(R^1)}_l$
		4	2-Cl	H	O	Br	H	O	4-F
		4	2-Cl	H	O	F	H	O	4-F
		4	2-Cl	H	O	CH ₃	H	O	4-F
		4	2-Cl	H	O	Cl	H	O	2-F
		4	2-Cl	H	O	Br	H	O	2-F
		4	2-Cl	H	O	F	H	O	2-F
		4	2-Cl	H	O	CH ₃	H	O	2-F
		4	2-Cl	H	O	Cl	H	O	2-Cl
		4	2-Cl	H	O	Br	H	O	2-Cl
		4	2-Cl	H	O	F	H	O	2-Cl
		4	2-Cl	H	O	CH ₃	H	O	2-Cl
		4	2-Cl	H	O	Cl	H	O	3-Cl
		4	2-Cl	H	O	Br	H	O	3-Cl
		4	2-Cl	H	O	F	H	O	3-Cl
		4	2-Cl	H	O	CH ₃	H	O	3-Cl
		4	2-Cl	H	O	Cl	H	O	4-Cl

5	(continued)	\underline{A}	\underline{B}
10		$\frac{(R^5)}{(R^5)}_n$	
15			
20		$\underline{R^4}$	
25		\underline{Y}	
30		$\underline{R^3}$	
35		$\underline{R^2}_m$	
40		\underline{X}	
45			
50		$\underline{(R^1)}_l$	
55			

5	(continued)	\underline{A}	\underline{B}
10		$\frac{(R^5)}{(R^5)_n}$	
15			
20		$\underline{R^4}$	
25		\underline{Y}	
30		$\underline{R^3}$	
35		$\underline{R^2}$	
40		\underline{X}	
45			
50		$\underline{(R^1)_1}$	
55			

	$\frac{(R^1)_1}{1}$	$\frac{X}{1}$	$\frac{R^2}{m}$	$\frac{R^3}{1}$	$\frac{Y}{1}$	$\frac{R^4}{1}$	$\frac{(R^5)_n}{n}$	$\frac{A}{1}$	$\frac{B}{1}$
	3-CF ₃	O	H	Cl	O	H	2-Cl	4	5
	3-CF ₃	O	H	Br	O	H	2-Cl	4	5
	3-CF ₃	O	H	F	O	H	2-Cl	4	5
	3-CF ₃	O	H	CH ₃	O	H	2-Cl	4	5
	2-C ₂ H ₅	O	H	Cl	O	H	2-Cl	4	5
	2-NO ₂	O	H	Cl	O	H	2-Cl	4	5
	2-CF ₃	O	H	Cl	O	H	2-Cl	4	5
	3-OCH ₃	O	H	Cl	O	H	2-Cl	4	5
	4-OCHF ₂	O	H	Cl	O	H	2-Cl	4	5
	4-C ₂ H ₅	O	H	Cl	O	H	2-Cl	4	5
	3-CN	O	H	Cl	O	H	2-Cl	4	5
	4-CF ₃	O	H	Cl	O	H	2-Cl	4	5
	4-CF ₃	O	H	Br	O	H	2-Cl	4	5
	4-CF ₃	O	H	F	O	H	2-Cl	4	5
	4-CF ₃	O	H	CH ₃	O	H	2-Cl	4	5
	4-(iso)C ₃ H ₇	O	H	Cl	O	H	2-Cl	4	5
	4-(n)C ₄ H ₉	O	H	Cl	O	H	2-Cl	4	5

	5	10	15	20	25	30	35	40	45	50	55
	(continued)										
	\underline{A}	$\underline{(R^5)_n}$		$\underline{R^4}$	\underline{Y}	$\underline{R^3}$	$\underline{R^2_m}$	\underline{X}		$\underline{(R^1)_1}$	
	4	2-Cl		H	O	Cl	H	O		4-OCH ₃	
	4	2-Cl		H	O	Cl	H	O		3-NO ₂	
	4	2-Cl		H	O	Cl	H	O		4-CN	
	4	2-Cl		H	O	Cl	H	O		4-NO ₂	
	4	2-Cl		H	O	Cl	H	>C=O		H	
	4	2-Cl		H	O	Cl	H	NH		H	
	4	2-Cl		H	O	Cl	H	S		H	
	4	2-Cl		H	O	Cl	H	>S=O		H	
	4	2-Cl		H	O	Cl	H	>S(=O) ₂		H	
	4	2-Cl		H	O	Cl	H	CH ₂		H	
	4	2-CH ₃		H	O	Cl	H	CH ₂		H	
	4	2-Cl		H	S	Cl	H	O		H	
	4	2-CH ₃		H	S	Cl	H	O		H	
	4	2,6-Cl ₂		H	O	Cl	H	O		H	
	4	3-Cl		H	O	Cl	H	O		H	

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(continued)

$\frac{(R^1)_1}{H}$	\underline{X}	$\frac{R^2}{H}$	$\frac{R^3}{Cl}$	\underline{Y}	$\frac{R^4}{H}$	$\frac{(R^5)_n}{2,6-Cl_2}$	\underline{A}	\underline{B}
H	O	H	Cl	O	H	2,6-Cl ₂	4	5
H	O	H	Cl	O	H	2-CH ₃	4	5
H	O	H	Br	O	H	2-CH ₃	4	5
H	O	H	F	O	H	2-CH ₃	4	5
H	O	H	CH ₃	O	H	2-CH ₃	4	5
H	O	H	C ₂ H ₅	O	H	2-CH ₃	4	5
H	O	H	(n)C ₃ H ₇	O	H	2-CH ₃	4	5
3-F	O	H	Cl	O	H	2-CH ₃	4	5
3-F	O	H	Br	O	H	2-CH ₃	4	5
3-F	O	H	F	O	H	2-CH ₃	4	5
3-F	O	H	CH ₃	O	H	2-CH ₃	4	5
4-F	O	H	Cl	O	H	2-CH ₃	4	5
4-F	O	H	Br	O	H	2-CH ₃	4	5
4-F	O	H	F	O	H	2-CH ₃	4	5
4-F	O	H	CH ₃	O	H	2-CH ₃	4	5
2-F	O	H	Cl	O	H	2-CH ₃	4	5
3-Cl	O	H	Cl	O	H	2-CH ₃	4	5

	50	45	40	35	30	25	20	15	10	5
	$\frac{(R^1)_1}{1}$		\underline{X}	$\frac{R^2}{n}$	$\frac{R^3}{1}$	\underline{Y}	$\frac{R^4}{1}$	(continued)	$\frac{(R^5)_n}{1}$	$\frac{A}{1}$ $\frac{B}{1}$
	3-Cl		O	H	Br	O	H		2-CH ₃	4 5
	3-Cl		O	H	F	O	H		2-CH ₃	4 5
	3-Cl		O	H	CH ₃	O	H		2-CH ₃	4 5
	4-Cl		O	H	Cl	O	H		2-CH ₃	4 5
	4-Cl		O	H	Br	O	H		2-CH ₃	4 5
	4-Cl		O	H	F	O	H		2-CH ₃	4 5
	4-Cl		O	H	CH ₃	O	H		2-CH ₃	4 5
	2-Cl		O	H	Cl	O	H		2-CH ₃	4 5
	2-Br		O	H	Cl	O	H		2-CH ₃	4 5
	3-Br		O	H	Cl	O	H		2-CH ₃	4 5
	4-Br		O	H	Cl	O	H		2-CH ₃	4 5
	4-I		O	H	Cl	O	H		2-CH ₃	4 5
	4-I		O	H	Br	O	H		2-CH ₃	4 5
	4-I		O	H	F	O	H		2-CH ₃	4 5
	4-I		O	H	CH ₃	O	H		2-CH ₃	4 5
	3-I		O	H	Cl	O	H		2-CH ₃	4 5
	3-I		O	H	Br	O	H		2-CH ₃	4 5

5		<u>A</u>	5	<u>B</u>	5
10	(continued)	$\frac{(R^5)_n}{2-Cl}$	2-Cl		
15					
20		$\frac{R^4}{H}$	H		
25		<u>Y</u>	O		
30		$\frac{R^3}{Cl}$	Cl		
35		$\frac{R^2}{H}$	H		
40		<u>X</u>	O		
45					
50		$\frac{(R^1)_1}{H}$	H		
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(continued)

$\frac{(R^1)_1}{1}$	$\frac{X}{1}$	$\frac{R^2}{m}$	$\frac{R^3}{1}$	$\frac{Y}{1}$	$\frac{R^4}{1}$	$\frac{(R^5)_n}{1}$	$\frac{A}{1}$	$\frac{B}{1}$
4-Br	O	H	Cl	O	H	2-Cl	5	5
3-Br	O	H	Cl	O	H	2-Cl	5	5
2-Br	O	H	Cl	O	H	2-Cl	5	5
3-CH ₃	O	H	Cl	O	H	2-Cl	5	5
4-CH ₃	O	H	Cl	O	H	2-Cl	5	5
3-CF ₃	O	H	Cl	O	H	2-Cl	5	5
2-CH ₃	O	H	Cl	O	H	2-Cl	5	5
3-I	O	H	Cl	O	H	2-Cl	5	5
4-I	O	H	Cl	O	H	2-Cl	5	5
3-OCH ₃	O	H	Cl	O	H	2-Cl	5	5
4-OCHF ₂	O	H	Cl	O	H	2-Cl	5	5
4-CF ₃	O	H	Cl	O	H	2-Cl	5	5
H	O	H	Cl	O	H	2-CH ₃	5	5
3-F	O	H	Cl	O	H	2-CH ₃	5	5
3-F	O	H	Br	O	H	2-CH ₃	5	5
3-F	O	H	F	O	H	2-CH ₃	5	5
3-F	O	H	CH ₃	O	H	2-CH ₃	5	5

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(continued)

$(R^1)_1$	\underline{X}	$\underline{R^2_m}$	$\underline{R^3}$	\underline{Y}	$\underline{R^4}$	$\underline{(R^5)_n}$	\underline{A}	\underline{B}
3-Cl	O	H	Cl	O	H	2-CH ₃	5	5
4-Cl	O	H	Cl	O	H	2-CH ₃	5	5
3,5-F ₂	O	H	Cl	O	H	2-Cl	4	5
3,5-F ₂	O	H	Br	O	H	2-Cl	4	5
3,5-F ₂	O	H	F	O	H	2-Cl	4	5
3,5-F ₂	O	H	CH ₃	O	H	2-Cl	4	5
2,4-F ₂	O	H	Cl	O	H	2-Cl	4	5
2,4-F ₂	O	H	Br	O	H	2-Cl	4	5
2,4-F ₂	O	H	F	O	H	2-Cl	4	5
2,4-F ₂	O	H	CH ₃	O	H	2-Cl	4	5
3,4-F ₂	O	H	Cl	O	H	2-Cl	4	5
3,4-F ₂	O	H	Br	O	H	2-Cl	4	5
3,4-F ₂	O	H	F	O	H	2-Cl	4	5
3,4-F ₂	O	H	CH ₃	O	H	2-Cl	4	5
2,6-F ₂	O	H	Cl	O	H	2-Cl	4	5
2,6-F ₂	O	H	F	O	H	2-Cl	4	5

$\frac{(R^1)_1}{X}$	$\frac{R^2}{R^m}$	$\frac{R^3}{Y}$	$\frac{R^4}{R^n}$	$\frac{(R^5)_n}{A}$	$\frac{R^6}{B}$
0	H	Br	H	2-Cl	4
2,6-F ₂					5
0	H	CH ₃	H	2-Cl	4
2,6-F ₂					5
0	H	Cl	H	2-Cl	4
2,5-F ₂					5
0	H	Br	H	2-Cl	4
2,5-F ₂					5
0	H	F	H	2-Cl	4
2,5-F ₂					5
0	H	CH ₃	H	2-Cl	4
2,5-F ₂					5
0	H	Cl	H	2-Cl	4
2,3-F ₂					5
0	H	Br	H	2-Cl	4
2,3-F ₂					5
0	H	F	H	2-Cl	4
2,3-F ₂					5
0	H	CH ₃	H	2-Cl	4
2,3-F ₂					5
0	H	Cl	H	2-Cl	4
3,4-Cl ₂					5
0	H	Br	H	2-Cl	4
3,4-Cl ₂					5
0	H	F	H	2-Cl	4
3,4-Cl ₂					5
0	H	CH ₃	H	2-Cl	4
3,4-Cl ₂					5
0	H	Cl	H	2-Cl	4
3,5-Cl ₂					5
0	H	Br	H	2-Cl	4
3,5-Cl ₂					5

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$\frac{(R^1)_1}{1}$	\underline{X}	$\frac{R^2}{m}$	$\frac{R^3}{3}$	\underline{Y}	$\frac{R^4}{4}$	$\frac{(R^5)_n}{n}$	\underline{A}	\underline{B}
3,5-Br ₂	O	H	CH ₃	O	H	2-Cl	4	5
2,3-Br ₂	O	H	CH ₃	O	H	2-Cl	4	5
2,3-Br ₂	O	H	Cl	O	H	2-Cl	4	5
2,3-Br ₂	O	H	Br	O	H	2-Cl	4	5
2,3-Br ₂	O	H	F	O	H	2-Cl	4	5
2,5-Br ₂	O	H	Cl	O	H	2-Cl	4	5
2,5-Br ₂	O	H	Br	O	H	2-Cl	4	5
2,5-Br ₂	O	H	F	O	H	2-Cl	4	5
2,5-Br ₂	O	H	CH ₃	O	H	2-Cl	4	5
3,4-(CH ₃) ₂	O	H	Cl	O	H	2-Cl	4	5
3,4-(CH ₃) ₂	O	H	F	O	H	2-Cl	4	5
3,4-(CH ₃) ₂	O	H	Br	O	H	2-Cl	4	5
3,4-(CH ₃) ₂	O	H	CH ₃	O	H	2-Cl	4	5
3,5-(CH ₃) ₂	O	H	Cl	O	H	2-Cl	4	5
3,5-(CH ₃) ₂	O	H	Br	O	H	2-Cl	4	5
3,5-(CH ₃) ₂	O	H	F	O	H	2-Cl	4	5

[illegible]

$\frac{(R^1)_1}{X}$	$\frac{R^2}{R^m}$	$\frac{R^3}{Y}$	$\frac{R^4}{R^n}$	$\frac{(R^5)_n}{A}$	$\frac{B}{C}$
3,4,5-Cl ₃	H	CH ₃	H	2-Cl	4 5
2-F, 4-Cl	H	Cl	H	2-Cl	4 5
2-F, 4-Cl	H	F	H	2-Cl	4 5
2-F, 4-Cl	H	Br	H	2-Cl	4 5
2-F, 4-Cl	H	CH ₃	H	2-Cl	4 5
2-Cl, 4-Br	H	Cl	H	2-Cl	4 5
2-Cl, 4-Br	H	F	H	2-Cl	4 5
2-Cl, 4-Br	H	Br	H	2-Cl	4 5
2-Cl, 4-Br	H	CH ₃	H	2-Cl	4 5
2-Cl, 4-CH ₃	H	Cl	H	2-Cl	4 5
2-Cl, 4-CH ₃	H	F	H	2-Cl	4 5
2-Cl, 4-CH ₃	H	Br	H	2-Cl	4 5
2-Cl, 4-CH ₃	H	CH ₃	H	2-Cl	4 5
2-F, 4-CH ₃	H	Cl	H	2-Cl	4 5
2-F, 4-CH ₃	H	F	H	2-Cl	4 5
2-F, 4-CH ₃	H	Br	H	2-Cl	4 5

$\frac{(R^1)_1}{X}$	$\frac{R^2}{R^m}$	$\frac{R^3}{R}$	$\frac{Y}{R^4}$	$\frac{(R^5)_n}{A}$	$\frac{B}{B}$
2-F, 4-CH ₃	H	CH ₃	O	2-Cl	4
2,4-Br ₂	H	Cl	O	2-Cl	4
2,4-Br ₂	H	F	O	2-Cl	4
2,4-Br ₂	H	Br	O	2-Cl	4
2,4-Br ₂	H	CH ₃	O	2-Cl	4
2-F, 4-Br	H	Cl	O	2-Cl	4
2-F, 4-Br	H	F	O	2-Cl	4
2-F, 4-Br	H	Br	O	2-Cl	4
2-F, 4-Br	H	CH ₃	O	2-Cl	4
2-Cl, 4-CF ₃	H	Cl	O	2-Cl	4
2-Cl, 4-CF ₃	H	F	O	2-Cl	4
2-Cl, 4-CF ₃	H	Br	O	2-Cl	4
2-Cl, 4-CF ₃	H	CH ₃	O	2-Cl	4
2-Cl, 4-F	H	Cl	O	2-Cl	4
2-Cl, 4-F	H	F	O	2-Cl	4
2-Cl, 4-F	H	Br	O	2-Cl	4

$\frac{(R^1)_1}{X}$	$\frac{R^2}{R^m}$	$\frac{R^3}{R^3}$	$\frac{R^4}{R^4}$	$\frac{(R^5)_n}{A}$	$\frac{R^6}{B}$
2-Cl, 4-F	H	CH ₃	H	2-Cl	4
2-CF ₃ , 4-Cl	H	Cl	H	2-Cl	4
2-CF ₃ , 4-Cl	H	F	H	2-Cl	4
2-CF ₃ , 4-Cl	H	Br	H	2-Cl	4
2-CF ₃ , 4-Cl	H	CH ₃	H	2-Cl	4
2-CH ₃ , 4-Cl	H	F	H	2-Cl	4
2-CH ₃ , 4-Cl	H	Cl	H	2-Cl	4
2-CH ₃ , 4-Cl	H	Br	H	2-Cl	4
2-CH ₃ , 4-Cl	H	CH ₃	H	2-Cl	4
2-Br, 4-CH ₃	H	Cl	H	2-Cl	4
2-Br, 4-CH ₃	H	F	H	2-Cl	4
2-Br, 4-CH ₃	H	Br	H	2-Cl	4
2-Br, 4-CH ₃	H	CH ₃	H	2-Cl	4
3-F, 4-CH ₃	H	Cl	H	2-Cl	4
3-F, 4-CH ₃	H	F	H	2-Cl	4
3-F, 4-CH ₃	H	Br	H	2-Cl	4

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$\frac{(R^1)_1}{(R^1)_1}$	\underline{X}	$\frac{R^2}{R^2-m}$	$\frac{R^3}{R^3}$	\underline{Y}	$\frac{R^4}{R^4}$	$\frac{(R^5)_n}{(R^5)_n}$	\underline{A}	\underline{B}
3-CH ₃ , 4-Br	O	H	CH ₃	O	H	2-Cl	4	5
3-Cl, 4-F	O	H	Cl	O	H	2-Cl	4	5
3-Cl, 4-F	O	H	F	O	H	2-Cl	4	5
3-Cl, 4-F	O	H	Br	O	H	2-Cl	4	5
3-Cl, 4-F	O	H	CH ₃	O	H	2-Cl	4	5
3-F, 4-Cl	O	H	Cl	O	H	2-Cl	4	5
3-F, 4-Cl	O	H	F	O	H	2-Cl	4	5
3-F, 4-Cl	O	H	Br	O	H	2-Cl	4	5
3-F, 4-Cl	O	H	CH ₃	O	H	2-Cl	4	5
3-F, 4-Br	O	H	Cl	O	H	2-Cl	4	5
3-F, 4-Br	O	H	F	O	H	2-Cl	4	5
3-F, 4-Br	O	H	Br	O	H	2-Cl	4	5
3-F, 4-Br	O	H	CH ₃	O	H	2-Cl	4	5
3-Cl, 4-Br	O	H	Cl	O	H	2-Cl	4	5
3-Cl, 4-Br	O	H	F	O	H	2-Cl	4	5
3-Cl, 4-Br	O	H	Br	O	H	2-Cl	4	5

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$\frac{(R^1)_1}{1}$	\underline{X}	$\frac{R^2}{m}$	$\frac{R^3}{1}$	\underline{Y}	$\frac{R^4}{1}$	$\frac{(R^5)_n}{n}$	\underline{A}	\underline{B}
3-Cl, 4-Br	O	H	CH ₃	O	H	2-Cl	4	5
3-CF ₃ , 4-Cl	O	H	Cl	O	H	2-Cl	4	5
3-CF ₃ , 4-Cl	O	H	F	O	H	2-Cl	4	5
3-CF ₃ , 4-Cl	O	H	Br	O	H	2-Cl	4	5
3-CF ₃ , 4-Cl	O	H	CH ₃	O	H	2-Cl	4	5
3-NO ₂ , 4-Cl	O	H	F	O	H	2-Cl	4	5
3-NO ₂ , 4-Cl	O	H	Cl	O	H	2-Cl	4	5
3-NO ₂ , 4-Cl	O	H	Br	O	H	2-Cl	4	5
3-NO ₂ , 4-Cl	O	H	CH ₃	O	H	2-Cl	4	5
2-Cl, 5-CH ₃	O	H	Cl	O	H	2-Cl	4	5
2-Cl, 5-CH ₃	O	H	F	O	H	2-Cl	4	5
2-Cl, 5-CH ₃	O	H	Br	O	H	2-Cl	4	5
2-Cl, 5-CH ₃	O	H	CH ₃	O	H	2-Cl	4	5
2-F, 5-CH ₃	O	H	Cl	O	H	2-Cl	4	5
2-F, 5-CH ₃	O	H	F	O	H	2-Cl	4	5
2-F, 5-CH ₃	O	H	Br	O	H	2-Cl	4	5
2-F, 5-CH ₃	O	H	CH ₃	O	H	2-Cl	4	5

[illegible]

$\frac{(R^1)_1}{1}$	\underline{X}	$\frac{R^2}{m}$	$\frac{R^3}{1}$	\underline{Y}	$\frac{R^4}{1}$	$\frac{(R^5)_n}{n}$	$\frac{A}{1}$	$\frac{B}{1}$
2,4,6-Cl ₃	O	H	Cl	O	H	2-Cl	4	5
2,4-Cl ₂ , 3,5-(CH ₃) ₂	O	H	Cl	O	H	2-Cl	4	5
2,3,5,6-F ₄	O	H	Cl	O	H	2-Cl	4	5
2,3,4,5,6-F ₅	O	H	Cl	O	H	2-Cl	4	5
3,5-F ₂	O	H	Cl	O	H	2-CH ₃	4	5
3,5-F ₂	O	H	F	O	H	2-CH ₃	4	5
3,5-F ₂	O	H	Br	O	H	2-CH ₃	4	5
3,5-F ₂	O	H	CH ₃	O	H	2-CH ₃	4	5
3,4-Cl ₂	O	H	F	O	H	2-CH ₃	4	5
3,4-Cl ₂	O	H	Cl	O	H	2-CH ₃	4	5
3,4-Cl ₂	O	H	Br	O	H	2-CH ₃	4	5
3,4-Cl ₂	O	H	CH ₃	O	H	2-CH ₃	4	5
3,5-Cl ₂	O	H	Cl	O	H	2-CH ₃	4	5
3,5-Cl ₂	O	H	F	O	H	2-CH ₃	4	5
3,5-Cl ₂	O	H	Br	O	H	2-CH ₃	4	5
3,5-Cl ₂	O	H	CH ₃	O	H	2-CH ₃	4	5

$\frac{(R^1)_1}{X}$	$\frac{R^2}{R^m}$	$\frac{R^3}{Y}$	$\frac{R^4}{\frac{(R^5)_n}{A}}$	$\frac{B}{5}$
3,5-Cl ₂	H	CH ₃	H	4
2,4-F ₂	H	Cl	H	4
2,4-F ₂	H	Br	H	4
2,4-F ₂	H	F	H	4
2,4-F ₂	H	CH ₃	H	4
3,4-F ₂	H	Cl	H	4
3,4-F ₂	H	F	H	4
3,4-F ₂	H	Br	H	4
3,4-F ₂	H	CH ₃	H	4
2,4-Cl ₂	H	Cl	H	4
2,4-Cl ₂	H	F	H	4
2,4-Cl ₂	H	Br	H	4
2,4-Cl ₂	H	CH ₃	H	4
3,5-(CH ₃) ₂	H	Cl	H	4
3,5-(CH ₃) ₂	H	F	H	4
3,5-(CH ₃) ₂	H	Br	H	4

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$(R^1)_1$	\underline{X}	$\underline{R^2_m}$	$\underline{R^3}$	\underline{Y}	$\underline{R^4}$	$\underline{(R^5)_n}$	\underline{A}	\underline{B}
3,5-(CH ₃) ₂	O	H	CH ₃	O	H	2-CH ₃	4	5
3,4-(CH ₃) ₂	O	H	Cl	O	H	2-CH ₃	4	5
3,4-(CH ₃) ₂	O	H	F	O	H	2-CH ₃	4	5
3,4-(CH ₃) ₂	O	H	Br	O	H	2-CH ₃	4	5
3,4-(CH ₃) ₂	O	H	CH ₃	O	H	2-CH ₃	4	5
3,5-Br ₂	O	H	Cl	O	H	2-CH ₃	4	5
2,6-F ₂	O	H	Cl	O	H	2-CH ₃	4	5
2,5-Cl ₂	O	H	Cl	O	H	2-CH ₃	4	5
2,5-F ₂	O	H	Cl	O	H	2-CH ₃	4	5
2,5-F ₂	O	H	F	O	H	2-CH ₃	4	5
2,5-F ₂	O	H	Br	O	H	2-CH ₃	4	5
2,5-F ₂	O	H	CH ₃	O	H	2-CH ₃	4	5
2,3-F ₂	O	H	Cl	O	H	2-CH ₃	4	5
2,3-F ₂	O	H	F	O	H	2-CH ₃	4	5
2,3-F ₂	O	H	Br	O	H	2-CH ₃	4	5
2,3-F ₂	O	H	CH ₃	O	H	2-CH ₃	4	5

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$\frac{(R^1)_1}{1}$	$\frac{X}{1}$	$\frac{R^2}{m}$	$\frac{R^3}{1}$	$\frac{Y}{1}$	$\frac{R^4}{1}$	$\frac{(R^5)_n}{n}$	$\frac{A}{1}$	$\frac{B}{1}$
2,5-(CH ₃) ₂	O	H	Cl	O	H	2-CH ₃	4	5
2,3-Cl ₂	O	H	Cl	O	H	2-CH ₃	4	5
2,3-(CH ₃) ₂	O	H	Cl	O	H	2-CH ₃	4	5
2,3-Br ₂	O	H	Cl	O	H	2-CH ₃	4	5
2,5-Br ₂	O	H	Cl	O	H	2-CH ₃	4	5
3,5-(CF ₃) ₂	O	H	Cl	O	H	2-CH ₃	4	5
3,4,5-Cl ₃	O	H	Cl	O	H	2-CH ₃	4	5
2-F, 4-Cl	O	H	Cl	O	H	2-CH ₃	4	5
2-Cl, 4-Br	O	H	Cl	O	H	2-CH ₃	4	5
2-Cl, 4-CH ₃	O	H	Cl	O	H	2-CH ₃	4	5
2-F, 4-CH ₃	O	H	Cl	O	H	2-CH ₃	4	5
2,4-Br ₂	O	H	Cl	O	H	2-CH ₃	4	5
2-F, 4-Br	O	H	Cl	O	H	2-CH ₃	4	5
2-Cl, 4-CF ₃	O	H	Cl	O	H	2-CH ₃	4	5
2-Cl, 4-F	O	H	Cl	O	H	2-CH ₃	4	5
2-CF ₃ , 4-Cl	O	H	Cl	O	H	2-CH ₃	4	5

\underline{X}	$\underline{(R^1)_1}$	$\underline{R^2_m}$	$\underline{R^3}$	\underline{Y}	$\underline{R^4}$	$\underline{(R^5)_n}$	\underline{A}	\underline{B}
O	2-CH ₃ , 4-Cl	H	Cl	O	H	2-CH ₃	4	5
O	2-Br, 4-CH ₃	H	Cl	O	H	2-CH ₃	4	5
O	3-F, 4-CH ₃	H	Cl	O	H	2-CH ₃	4	5
O	3-F, 4-CH ₃	H	F	O	H	2-CH ₃	4	5
O	3-F, 4-CH ₃	H	Br	O	H	2-CH ₃	4	5
O	3-F, 4-CH ₃	H	CH ₃	O	H	2-CH ₃	4	5
O	3-Cl, 4-CH ₃	H	F	O	H	2-CH ₃	4	5
O	3-Cl, 4-CH ₃	H	Cl	O	H	2-CH ₃	4	5
O	3-Cl, 4-CH ₃	H	Br	O	H	2-CH ₃	4	5
O	3-Cl, 4-CH ₃	H	CH ₃	O	H	2-CH ₃	4	5
O	3-Br, 4-CH ₃	H	Cl	O	H	2-CH ₃	4	5
O	3-CH ₃ , 4-F	H	Cl	O	H	2-CH ₃	4	5
O	3-CH ₃ , 4-Br	H	Cl	O	H	2-CH ₃	4	5
O	3-Cl, 4-F	H	Cl	O	H	2-CH ₃	4	5
O	3-Cl, 4-F	H	Br	O	H	2-CH ₃	4	5
O	3-Cl, 4-F	H	F	O	H	2-CH ₃	4	5

$\frac{(R^1)_1}{X}$	$\frac{R^2}{R^m}$	$\frac{R^3}{Y}$	$\frac{R^4}{\frac{(R^5)_n}{A}}$	$\frac{R^5}{B}$
3-Cl, 4-F	H	CH ₃	H	2-CH ₃
3-F, 4-Cl	H	Cl	H	2-CH ₃
3-F, 4-Cl	H	F	H	2-CH ₃
3-F, 4-Cl	H	Br	H	2-CH ₃
3-F, 4-Cl	H	CH ₃	H	2-CH ₃
3-F, 4-Br	H	Cl	H	2-CH ₃
3-F, 4-Br	H	F	H	2-CH ₃
3-F, 4-Br	H	Br	H	2-CH ₃
3-F, 4-Br	H	CH ₃	H	2-CH ₃
3-Cl, 4-Br	H	F	H	2-CH ₃
3-Cl, 4-Br	H	Cl	H	2-CH ₃
3-Cl, 4-Br	H	Br	H	2-CH ₃
3-Cl, 4-Br	H	CH ₃	H	2-CH ₃
3-CF ₃ , 4-Cl	H	Cl	H	2-CH ₃
2-Cl, 5-CH ₃	H	Cl	H	2-CH ₃
2-F, 5-CH ₃	H	Cl	H	2-CH ₃

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(continued)

$\frac{(R^1)_1}{1}$	\underline{X}	$\frac{R^2}{m}$	$\frac{R^3}{1}$	\underline{Y}	$\frac{R^4}{1}$	$\frac{(R^5)_n}{n}$	\underline{A}	\underline{B}
3-CH ₃ , 4-Cl	O	H	Cl	O	H	2-CH ₃	4	5
3-CH ₃ , 4-Cl	O	H	F	O	H	2-CH ₃	4	5
3-CH ₃ , 4-Cl	O	H	Br	O	H	2-CH ₃	4	5
3-CH ₃ , 4-Cl	O	H	CH ₃	O	H	2-CH ₃	4	5
3,5-(CH ₃) ₂ , 4-Cl	O	H	Cl	O	H	2-CH ₃	4	5
2,3,4-Cl ₃	O	H	Cl	O	H	2-CH ₃	4	5
2,4,5-Cl ₃	O	H	Cl	O	H	2-CH ₃	4	5
2,4,5-Cl ₃	O	H	F	O	H	2-CH ₃	4	5
2,4,5-Cl ₃	O	H	Br	O	H	2-CH ₃	4	5
2,4,5-Cl ₃	O	H	CH ₃	O	H	2-CH ₃	4	5
3,5-F ₂	O	H	Cl	O	H	2-CH ₃	5	5
3,4-Cl ₂	O	H	Cl	O	H	2-CH ₃	5	5
3,5-Cl ₂	O	H	Cl	O	H	2-CH ₃	5	5
2,4-F ₂	O	H	Cl	O	H	2-CH ₃	5	5
3,4-F ₂	O	H	Cl	O	H	2-CH ₃	5	5
2,4-Cl ₂	O	H	Cl	O	H	2-CH ₃	5	5

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$\frac{(R^1)_1}{1}$	\underline{X}	$\frac{R^2}{m}$	$\underline{R^3}$	\underline{Y}	$\frac{R^4}{H}$	$\frac{(R^5)_n}{n}$	\underline{A}	\underline{B}
3, 4-(CH ₃) ₂	O	H	Cl	O	H	2-CH ₃	5	5
3, 5-(CH ₃) ₂	O	H	Cl	O	H	2-CH ₃	5	5
3, 5-Br ₂	O	H	Cl	O	H	2-CH ₃	5	5
2, 6-F ₂	O	H	Cl	O	H	2-CH ₃	5	5
2, 5-Cl ₂	O	H	Cl	O	H	2-CH ₃	5	5
2, 5-F ₂	O	H	Cl	O	H	2-CH ₃	5	5
2, 5-(CH ₃) ₂	O	H	Cl	O	H	2-CH ₃	5	5
2, 3-F ₂	O	H	Cl	O	H	2-CH ₃	5	5
2, 3-Cl ₂	O	H	Cl	O	H	2-CH ₃	5	5
2, 3-(CH ₃) ₂	O	H	Cl	O	H	2-CH ₃	5	5
3, 4-Br ₂	O	H	Cl	O	H	2-CH ₃	5	5
2, 5-Br ₂	O	H	Cl	O	H	2-CH ₃	5	5
2-F, 4-Cl	O	H	Cl	O	H	2-CH ₃	5	5
2-Cl, 4-Br	O	H	Cl	O	H	2-CH ₃	5	5
2-Cl, 4-CH ₃	O	H	Cl	O	H	2-CH ₃	5	5
3-F, 4-CH ₃	O	H	Cl	O	H	2-CH ₃	5	5

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	(continued)									
	$\frac{(R^1)_1}{2, 4-Br_2}$	$\frac{X}{O}$	$\frac{R^2_m}{H}$	$\frac{R^3}{Cl}$	$\frac{Y}{O}$	$\frac{R^4}{H}$	$\frac{(R^5)_n}{2-CH_3}$	$\frac{A}{5}$	$\frac{B}{5}$	
	2-F, 4-Br	O	H	Cl	O	H	2-CH ₃	5	5	
	2-Cl, 4-CF ₃	O	H	Cl	O	H	2-CH ₃	5	5	
	2-Cl, 4-F	O	H	Cl	O	H	2-CH ₃	5	5	
	2-CF ₃ , 4-Cl	O	H	Cl	O	H	2-CH ₃	5	5	
	2-CH ₃ , 4-Cl	O	H	Cl	O	H	2-CH ₃	5	5	
	2-Br, 4-CH ₃	O	H	Cl	O	H	2-CH ₃	5	5	
	3-F, 4-CH ₃	O	H	Cl	O	H	2-CH ₃	5	5	
	3-Cl, 4-CH ₃	O	H	Cl	O	H	2-CH ₃	5	5	
	3-Br, 4-CH ₃	O	H	Cl	O	H	2-CH ₃	5	5	
	3-CH ₃ , 4-F	O	H	Cl	O	H	2-CH ₃	5	5	
	3-CH ₃ , 4-Br	O	H	Cl	O	H	2-CH ₃	5	5	
	3-Cl, 4-F	O	H	Cl	O	H	2-CH ₃	5	5	
	3-F, 4-Cl	O	H	Cl	O	H	2-CH ₃	5	5	
	3-F, 4-Br	O	H	Cl	O	H	2-CH ₃	5	5	
	3-Cl, 4-Br	O	H	Cl	O	H	2-CH ₃	5	5	
	3-Cl, 4-Br	O	H	Cl	O	H	2-CH ₃	5	5	

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$\frac{(R^1)_1}{1}$	\underline{X}	$\frac{R^2}{m}$	$\frac{R^3}{1}$	\underline{Y}	$\frac{R^4}{1}$	$\frac{(R^5)_n}{n}$	\underline{A}	\underline{B}
3-CF ₃ , 4-Cl	O	H	Cl	O	H	2-CH ₃	5	5
2-Cl, 5-CH ₃	O	H	Cl	O	H	2-CH ₃	5	5
2-F, 5-CH ₃	O	H	Cl	O	H	2-CH ₃	5	5
3-CH ₃ , 4-Cl	O	H	Cl	O	H	2-CH ₃	5	5
3,5-(CH ₃) ₂ , 4-Cl	O	H	Cl	O	H	2-CH ₃	5	5
3,4,5-Cl ₃	O	H	Cl	O	H	2-CH ₃	5	5
2,4,5-Cl ₃	O	H	Cl	O	H	2-CH ₃	5	5
2,4,6-Cl ₃	O	H	Cl	O	H	2-CH ₃	5	5
H	O	6-Cl	Cl	O	H	2-Cl	4	5
H	O	6-F	Cl	O	H	2-Cl	4	5
H	O	3,4-Cl ₂	Cl	O	H	2-Cl	4	5
H	O	3,4-(CH ₃) ₂	Cl	O	H	2-Cl	4	5
H	O	3-Cl	Cl	O	H	2-Cl	4	5
H	O	H	Cl	O	CH ₃	2-Cl	4	5
H	O	H	Cl	O	C ₂ H ₅	2-Cl	4	5
H	O	H	Cl	O	H	2-F	4	5

$\frac{(R^1)_1}{1}$	\underline{X}	$\frac{R^2}{m}$	$\frac{R^3}{3}$	\underline{Y}	$\frac{R^4}{4}$	$\frac{(R^5)_n}{n}$	\underline{A}	\underline{B}
H	O	H	Cl	O	H	2-Br	4	5
H	O	H	Cl	O	H	2-CF ₃	4	5
H	O	H	Cl	O	H	2-C ₂ H ₅	4	5
H	O	H	Cl	O	H	2-(iso)-C ₃ H ₇	4	5
H	O	H	Cl	O	H	2-(n)-C ₃ H ₇	4	5
H	O	H	Cl	O	H	2-I	4	5
H	O	H	Cl	O	H	2-I	5	5
3-F	O	H	Cl	O	H	2-I	4	5
3,5-F ₂	O	H	Cl	O	H	2-I	4	5
3,4-Cl ₂	O	H	Cl	O	H	2-I	4	5
3-CH ₃	O	H	Cl	O	H	2-I	4	5
H	O	H	Cl	O	H	2-OCH ₃	4	5
H	O	H	Cl	O	H	2-OC ₂ H ₅	4	5
H	O	H	Cl	O	H	2-OCF ₃	4	5
H	O	H	Cl	O	H	2-OCHF ₂	4	5
H	O	5-Cl	Cl	O	H	2-Cl	4	5

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$\frac{(R^1)_1}{1}$	\underline{X}	$\frac{R^2}{m}$	$\frac{R^3}{1}$	\underline{Y}	$\frac{R^4}{1}$	$\frac{(R^5)_n}{n}$	\underline{A}	\underline{B}
3-F	0	5-Cl	Cl	0	H	2-Cl	4	5
3,5-F ₂	0	5-Cl	Cl	0	H	2-Cl	4	5
3-Cl	0	5-Cl	Cl	0	H	2-Cl	4	5
3,4-Cl ₂	0	5-Cl	Cl	0	H	2-Cl	4	5
4-Cl	0	5-Cl	Cl	0	H	2-Cl	4	5
3,5-Cl ₂	0	5-Cl	Cl	0	H	2-Cl	4	5
2,4-F ₂	0	5-Cl	Cl	0	H	2-Cl	4	5
3,4-F ₂	0	5-Cl	Cl	0	H	2-Cl	4	5
4-F	0	5-Cl	Cl	0	H	2-Cl	4	5
3-CH ₃	0	5-Cl	Cl	0	H	2-Cl	4	5
3,4-(CH ₃) ₂	0	5-Cl	Cl	0	H	2-Cl	4	5
3-F, 4-CH ₃	0	5-Cl	Cl	0	H	2-Cl	4	5
3-F, 4-Cl	0	5-Cl	Cl	0	H	2-Cl	4	5
3-CH ₃ , 4-Cl	0	5-Cl	Cl	0	H	2-Cl	4	5
2,5-F ₂	0	5-Cl	Cl	0	H	2-Cl	4	5
3,4-F ₂	0	5-Cl	Cl	0	H	2-Cl	4	5
3-CH ₃ , 4-Cl	0	5-Cl	Cl	0	H	2-Cl	4	5

[illegible]

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(continued)

$\frac{(R^1)_1}{1}$	$\frac{X}{1}$	$\frac{R^2}{m}$	$\frac{R^3}{1}$	$\frac{Y}{1}$	$\frac{R^4}{1}$	$\frac{(R^5)_n}{n}$	$\frac{A}{1}$	$\frac{B}{1}$
3-F, 4-Cl	O	5-F	Cl	O	H	2-Cl	4	5
3-CH ₃ , 4-Cl	O	5-F	Cl	O	H	2-Cl	4	5
2,5-F ₂	O	5-F	Cl	O	H	2-Cl	4	5
H	O	5-CH ₃	Cl	O	H	2-Cl	4	5
3-F	O	5-CH ₃	Cl	O	H	2-Cl	4	5
3,5-F ₂	O	5-CH ₃	Cl	O	H	2-Cl	4	5
3-Cl	O	5-CH ₃	Cl	O	H	2-Cl	4	5
3,4-Cl ₂	O	5-CH ₃	Cl	O	H	2-Cl	4	5
4-Cl	O	5-CH ₃	Cl	O	H	2-Cl	4	5
3,5-Cl ₂	O	5-CH ₃	Cl	O	H	2-Cl	4	5
2,4-F ₂	O	5-CH ₃	Cl	O	H	2-Cl	4	5
3,4-F ₂	O	5-CH ₃	Cl	O	H	2-Cl	4	5
4-F	O	5-CH ₃	Cl	O	H	2-Cl	4	5
3-CH ₃	O	5-CH ₃	Cl	O	H	2-Cl	4	5
3,4-(CH ₃) ₂	O	5-CH ₃	Cl	O	H	2-Cl	4	5
3,5-(CH ₃) ₂	O	5-CH ₃	Cl	O	H	2-Cl	4	5

$(R^1)_1$	\underline{X}	$\underline{R^2_m}$	$\underline{R^3}$	\underline{Y}	$\underline{R^4}$	$\frac{(R^5)_n}{(R^5)}$	\underline{A}	\underline{B}
3-F, 4-CH ₃	O	5-CH ₃	Cl	O	H	2-Cl	4	5
3-Cl, 4-CH ₃	O	5-CH ₃	Cl	O	H	2-Cl	4	5
3-F, 4-Cl	O	5-CH ₃	Cl	O	H	2-Cl	4	5
3-CH ₃ , 4-Cl	O	5-CH ₃	Cl	O	H	2-Cl	4	5
2,5-F ₂	O	5-CH ₃	Cl	O	H	2-Cl	4	5
2,3-F ₂	O	5-CH ₃	Cl	O	H	2-Cl	4	5
3-CH ₃ , 4-Cl	O	5-CH ₃	Cl	O	H	2-Cl	4	5
3,5-F ₂	O	H	Cl	O	CH ₃	2-Cl	4	5
3,5-F ₂	O	H	Cl	O	C ₂ H ₅	2-Cl	4	5
3,4-Cl ₂	O	H	Cl	O	CH ₃	2-Cl	4	5
3-F	O	H	Cl	O	H	2-Cl	4	5
3,4-F ₂	O	H	Cl	O	(iso)C ₃ H ₇	2-Cl	4	5

Examples of the insect pests against which the pyridine derivatives (I) of the invention exhibit controlling
55 effects are as shown below.

Hemiptera:-

Planthoppers such as brown planthopper (Nilaparvata lugens), white-backed rice planthopper (Sogatella furcifera) and small brown planthopper (Laodelphax striatellus); leafhoppers such as green rice leafhopper (Nephotettix cincticeps), Nephotettix virescens, Nephotettix nigropictus, zig-zag rice leafhopper (Recilia dorsalis), tea green leafhopper (Empoasca onukii) and grape leafhopper (Arboridia apicalis); aphids such as cotton aphid (Aphis gossypii) and green peach aphid (Myzus persicae); bugs; whiteflies (Aleyrodidae) such as sweet potato whitefly (Bemisia tabaci) and greenhouse whitefly (Trialeurodes vaporariorum); scales; mealy bugs; lace bugs (Tingidae); psyllids (Psyllidae), etc.

Lepidoptera:-

Pyralid moths (Pyralidae) such as rice stem borer (Chilo suppressalis), rice leafroller (Cnaphalocrocis medinalis) and Indian meal moth (Plodia interpunctella); Noctuidae such as tobacco curworm (Spodoptera litura), rice armyworm (Pseudaletia separate), cabbage armyworm (Mamestra brassicae) and beet semi-looper (Autographa nigrisigna); Agrothis spp. such as turnip cutworm (Agrothis segetum) and black cutworm (Agrothis ipsilon); Heliothis spp.; Pieridae such as common cabbageworm (Pieris rapae crucivora); tortricid moths (Tortricidae) such as Adoxophyes spp. and Grapholita spp.; Carposinidae such as Lyonetiid moths (Lyonetiidae), leafblotch miners (Gracillariidae), gelechiid moths (Gelechiidae) and tussock moths (Lymantriidae); diamondback moth (Plutella xylostella), clothes moths (Tineidae), casemaking clothes moth (Tinea translucens) and webbing clothes moth (Tineola bisselliella), etc.

Diptera:-

Mosquitos (Calicidae) such as common mosquito (Culex pipiens pallens) and Culex tritaeniorhynchus; Aedes spp. such as Aedes aegypti and Aedes albopictus; Anopheles spp. such as Anopheles sinensis; midges (Chironomidae); Muscidae such as housefly (Musca domestica) and false stablefly (Muscina stabulans); Calliphoridae; Sarcophagidae; lesser housefly (Fannia canicularis); anthomyiid flies (Anthomyiidae) such as seedcorn maggot (Delia platura) and onion maggot (Delia antiqua); fruit flies (Tephritidae); shore flies (Ephydriidae); small fruit flies (Drosophilidae); moth flies (Psychodidae); black flies (Simuliidae); Tabanidae; stable flies (Stomoxys); etc.

Coleoptera:-

Leaf beetles (Chrysomelidae) such as cucurbit beetle (Aulacophora femoralis), striped flea beetles (Phyllotreta striolata), western corn rootworm (Diabrotica virgifera) and southern corn root worm (Diabrotica undecimpunctata); scarabs (Scarabaeidae) such as cupreous chafer (Anomala cuprea) and soybean beetle (Anomala rufocuprea); weevils (Cureulionidae) such as maize weevil (Sitophilus zeamais), rice water weevil (Lissorhoptrus oryzophilus) and adzuki bean weevil (Callosobruchys chinensis), etc.; darkling beetles (Tenebrionidae) such as yellow mealworm (Tenebrio molitor) and red flour beetles (Tribolium castaneum); Anobiidae; Coccinellidae such as twenty-eight-spotted ladybirds (Epilachna vigintioctopunctata); powderpost beetles (Lyctidae); false powderpost beetles (Bostrychidae); Cerambycidae, etc.

Dictyoptera:-

Blattellidae such as German cockroach (Blattella germanica); Blattidae such as smokybrown cockroach (Periplaneta fuliginosa), American cockroach (Periplaneta americana), brown cockroach (Periplaneta brunnea) and oriental cockroach (Blatta orientalis), etc.

Thysanoptera:-

Thrips such as Thrips palmi, yellow tea thrips (Scirtothrips dorsalis) and flower thrips (Thrips hawaiiensis), etc.

Hymenoptera:-

Ants (Formicidae); sawflies (Tenthredinidae) such as cabbage sawfly (Athalia rosae ruficornis), etc.

Orthoptera:-

Mole crickets (Gryllotalpidae); grasshoppers (Acrididae), etc.

Aphaniptera:-

5 Purex irritans, etc.

Anoplura:-

10 Pediculus humanus capitis, Phthirus pubis, etc.

Isoptera:-

Reticulitermes speratus, Formosan subterranean termite (Coptotermes formosanus), etc.

15 Among the insect pests as above exemplified, the pyridine derivatives (I) are particularly effective in controlling those belonging to Hemiptera and also exhibit a remarkable controlling effect on planthoppers and leafhoppers in a field of rice plants.

The pyridine derivatives (I) may be used alone as insecticides or in mixtures with other insecticides and/or acaricides to enhance or expand their insecticidal or pesticidal use.

Examples of the other insecticides and acaricides are carbamate derivatives such as BPMC (2-sec-butylphenyl methylcarbamate), benfuracarb (ethyl N-[2,3-dihydro-2,2-dimethylbenzofuran-7-yloxy-carbonyl-(methyl)aminothio]-N-isopropyl-beta-alaninate), propoxur (2-isopropoxyphenyl N-methylcarbamate), carbosulfan (2,3-dihydro-2,2-dimethyl-7-benzofuran-5-yl N-methylcarbamate), carbaryl (1-naphthyl-N-methylcarbamate), methomyl (S-methyl-N-[(methylcarbamoyl)-oxy]thioacetimidate), ethiofencarb (2-(ethylthiomethyl)-phenyl methylcarbamate), aldicarb (2-methyl-2-(methylthio)-propionaldehyde O-methylcarbamoyloxime) and oxamyl (N,N-dimethyl-2-methylcarbamoyloxymino-2-(methylthio)acetamide); pyrethroids such as ethofenprop (2-(4-ethoxyphenyl)-2-methylpropyl-3-phenoxybenzyl ether), fenvalerate ((RS)-alpha-cyano-3-phenoxybenzyl (RS)-2-(4-chlorophenyl)-3-methylbutyrate), esfenvalerate ((S)-alpha-cyano-3-phenoxybenzyl (S)-2-(4-chlorophenyl)-3-methylbutyrate), fenpropathrin ((RS)-alpha-cyano-3-phenoxybenzyl 2,2,3,3-tetramethylcyclopropanecarboxylate), cypermethrin ((RS)-alpha-cyano-3-phenoxybenzyl (1RS,3RS)-(1RS,3RS)-3-(2,2-dichlorovinyl)-2,2-dimethylcyclopropanecarboxylate), permethrin (3-phenoxybenzyl (1RS,3RS)-(1RS,3RS)-3-(2,2-dichlorovinyl)-2,2-dimethylcyclopropanecarboxylate), cyhalothrin ((R,S)-alpha-cyano-3-phenoxybenzyl (Z)-(1RS,3RS)-3-(2-chloro-3,3,3-trifluoropropenyl)-2,2-dimethylcyclopropanecarboxylate), deltamethrin ((S)-alpha-cyano-m-phenoxybenzyl (1R,3R)-3-(2,2-dibromovinyl)-2-dimethylcyclopropanecarboxylate) and cycloprothrin ((RS)-alpha-cyano-3-phenoxybenzyl (RS)-2,2-dichloro-1-(4-ethoxyphenyl)-cyclopropanecarboxylate); thiadiazine derivatives such as buprofezin (2-tert-butylimino-3-isopropyl-5-phenyl-1,3,5-triazine-4-one); nitroimidazolidine derivatives such as imidacloprid (1-(6-chloro-3-pyridylmethyl)-N-nitroimidazolidine-2-ylideneamine); nereistoxin derivatives such as cartap (S,S'-(2-dimethylaminotrimethylene)bis(thiocarbamate)), thiocyclam (N,N-dimethyl-1,2,3-trithian-5-ylamine) and bensultap (S,S'-2-dimethylaminotrimethylene di(benzenethiosulphonate)); halogenated hydrocarbons such as endosulfan (6,7,8,9,10,10-hexachloro-1,5,5a,6,9,9a-hexahydro-6,9-methano-2,4,3-benzodioxathiepin-3-oxide) and gamma-BHC (1,2,3,4,5,6-hexachlorocyclohexane); benzoylphenylurea derivatives such as chlorfluazuron (1-[3,5-dichloro-4-(3-chloro-5-trifluoromethyl-pyridin-2-yloxy)phenyl]-3-(2,6-difluorobenzoyl)urea), teflubenzuron (1-(3,5-dichloro-2,4-difluorophenyl)-3-(2,6-difluorobenzoyl)urea) and flufenoxuron (1-[4-(2-chloro-4-trifluoromethylphenoxy)-2-fluorophenyl]-3-(2,6-difluorobenzoyl)urea); formamidine derivatives such as amitraz (N,N'-[(methylimino)dimethylidene]-di-2,4-xylydine) and chlordimeform (N'-(4-chloro-2-methylphenyl)-N,N-dimethylmethanimidamide), etc.

On the practical use of the pyridine derivatives (I) as insecticides, they may be employed as such but are normally mixed with appropriate additives such as solid carriers, liquid carriers, gaseous carriers, feed, etc. to formulate their compositions. When desired or necessary, surfactants and other adjuvants may be further incorporated therein. The compositions may be prepared into any conventional forms such as oil sprays, emulsifiable concentrates, wettable powders, flowable concentrates (e.g. water-based suspension formulations, water-based emulsion formulations), granules, dusts, aerosols, heating smoking formulations (e.g. self-burning-type smoking formulations, chemical reaction-type smoking formulations, porous ceramic plate-type smoking formulations), ULV formulations, poison baits, etc.

55 The composition of the invention contains generally the pyridine derivative(s) (I) as the active ingredient in an amount of from about 0.001 % to 95 % by weight based on the composition.

Examples of the solid carrier usable for making the composition are fine powders or granules of clays (e.g. kaolin clay, diatomaceous earth, synthetic hydrated silica, bentonite, Fubasami clay terra alba), talc,

ceramics, other inorganic minerals (e.g. sericite, quartz, sulfur, activated carbon, calcium carbonate, hydrated silica), chemical fertilizers (e.g. ammonium sulfate, ammonium phosphate, ammonium nitrate, urea, ammonium chloride), etc. Examples of the liquid carrier include water, alcohols (e.g. methanol, ethanol), ketones (e.g. acetone, methyl ethyl ketone), aromatic hydrocarbons (e.g. benzene, toluene, xylene, ethylbenzene, methylnaphthalene), aliphatic hydrocarbons (e.g. hexane, cyclohexane, kerosene, gas oil), esters (e.g. ethyl acetate, butyl acetate, etc.), nitriles (e.g. acetonitriles, isobutyronitrile), ethers (e.g. diisopropyl ether, dioxane), acid amides (e.g. N,N-dimethylformamide, N,N-dimethylacetamide), halogenated hydrocarbons (e.g. dichloromethane, trichloroethane, carbon tetrachloride), dimethyl sulfoxide, vegetable oils (e.g. soybean oil, cotton seed oil), etc. Examples of the gaseous carrier, i.e. a propellant, include freon gas, butane gas, LPG (liquefied petroleum gas), dimethyl ether, carbon dioxide, etc.

Examples of the surfactant are alkylsulfates, alkylsulfonates, alkylarylsulfonates, alkyl aryl ethers and polyoxyethylene derivatives thereof, polyethylene glycol ethers, polyvalent alcohol esters, sugar alcohol derivatives, etc. Examples of the adjuvants such as binders and dispersing agents are casein, gelatin, polysaccharides (e.g. starch powders, gum arabic, cellulose derivatives, alginic acid), lignin derivatives, bentonite, sugars, synthetic water-soluble high molecular weight substances (e.g. polyacrylic alcohol, polyvinylpyrrolidone, polyacrylic acid), etc. Examples of the stabilizer include PAP (acidic isopropyl phosphate), BHT (2,6-di-tert-butyl-4-methylphenol), BHA (mixture of 2-tert-butyl-4-methoxyphenol and 3-tert-butyl-4-methoxyphenol), vegetable oils, mineral oils, surfactants, fatty acids or esters thereof, etc.

The base material for self-burning-type smoking formulations may include, for example, burning heat-generating agents such as nitrates, nitrites, guanidine salts, potassium chlorate, nitrocellulose, ethyl cellulose and wood powders, pyrolysis-promoting agents such as alkali metal salts, alkaline earth metal salts, dichromates and chromates, oxygen-supplying agents such as potassium nitrate, burning-supporting agents such as melamine and wheat starch, extenders such as diatomaceous earth, binders such as synthetic pastes, etc. The base material for chemical reaction-type smoking formulations can include, for example, heat-generating agents such as alkali metal sulfides, alkali metal polysulfides, alkali metal hydrosulfides, hydrated salts of alkali metals and calcium oxide, catalyzing agents such as carbonaceous substances, iron carbide and activated clay, organic foaming agents such as azodicarbonamide, benzenesulfonyl hydrazides, dinitrosopentamethylenetetramine, polystyrene and polyurethane, fillers such as natural fiber pieces and synthetic fiber pieces, etc. The base material for poison baits may contain feed components such as crop powders, essential vegetable oil, sugars and crystalline cellulose, antioxidants such as dibutylhydroxytoluene and nordihydroguaiaretic acid, preservatives such as dehydroacetic acid, feeding error preventing agents such as red paper powders, incentive flavors such as cheese flavor and onion flavor, etc.

Flowable concentrates (water-based suspension or emulsion formulations) are generally obtained by dispersing about 1 to 75 parts by weight of the pyridine derivative (I) as the active ingredient finely and uniformly into water containing about 0.5 to 15 parts by weight of a dispersing agent, about 0.1 to 10 parts by weight of a suspending agent (e.g. protective colloids, compounds giving a thixotropic property) and optionally about 0 to 10 parts by weight of an auxiliary agent(s) such as a defoaming agent, an anticorrosive agent, a stabilizing agent, a spreading agent, penetration auxiliaries, antifreezing agent, an antibacterial agent, an antimolding agent and the like. The use of an oil, into which the active ingredient is hardly soluble, in place of water affords oil-based suspension formulations. Examples of the protective colloids as above mentioned are gelatin, casein, gums, cellulose ethers, polyvinyl alcohol, etc. Examples of the compounds giving a thixotropic property are bentonite, aluminum magnesium silicate, xanthane gum, polyacrylic acid, etc.

The composition of the invention thus obtained may be used as such or after diluting with water. It may be also used in a mixture with any other active component or composition chosen from insecticides, nematocides, acaricides, fungicides, bacteriocides, herbicides, plant growth regulators, synergistic agents, fertilizers, soil conditioners, animal feed, etc. Alternatively, the composition of the invention may be applied separately but simultaneously with said other active component or composition.

For the purpose of controlling insect pests in the agricultural field, the pyridine derivative (I) according to the present invention may be applied to the insect pests or the locus where the insect pests propagate generally in an amount of about 0.001 g to 500 g, and preferably about 0.1 g to 500 g per 10 ares. When the pyridine derivative (I) is applied in a form of emulsifiable concentrate, wettable powder, flowable concentrate or the like after dilution with water, its concentration may be from about 0.0001 to 1000 ppm. Granules, dusts, etc. may be used as such, i.e. without water dilution. When the pyridine derivative (I) is used for household or public hygiene, it may be used in the form of emulsifiable concentrate, wettable powder, flowable concentrate or the like with water dilution, etc. In this case, the concentration of the active ingredient may be from about 0.0001 to 10,000 ppm. In case of oils, aerosol, fumigants, ULV formulations,

poison baits, etc., they may be applied as such. However, the doses and concentrations may vary within broad ranges depending upon the composition, the application time, the place applied, the application method, the kind of insect pests, the condition of damage, etc. and may be increased or decreased, irrespective of the general ranges set forth above.

- 5 Practical and presently preferred embodiments of the invention will be hereinafter explained in more detail referring to Production Examples, Formulation Examples and Test Examples. These examples, however, should not be construed to be limitative.

In the following Production Examples, % is by weight unless otherwise indicated.

10 Production Example 1 (Production of Compound No. 1)

To a solution of 0.09 g of sodium hydride (60 % oil dispersion) in 10 ml of N,N-dimethylformamide, there was added dropwise a solution of 0.45 g of 2-chloro-4-phenoxyphenol in 3 ml of N,N-dimethylformamide with stirring and ice-cooling. After 30 minutes, a solution of 0.34 g of 2-chloro-5-chloromethylpyridine in 5 ml of N,N-dimethylformamide was added thereto at room temperature, followed by stirring at the same temperature for 10 hours. The reaction mixture was poured into ice-water and extracted twice with 100 ml of ethyl acetate. The extracts were combined together, washed with water, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. The residue was subjected to silica gel chromatography to give 0.50 g of 2-chloro-5-(2-chloro-4-phenoxyphenoxy)methylpyridine. Yield, 71 %. n_D^{22} : 1.6173

Production Example 2 (Production of Compound No. 33)

To a solution 0.07 g of sodium hydride (60 % oil dispersion) in 10 ml of N,N-dimethylformamide, there was added dropwise a solution of 0.50 g of 2-chloro-4-(3-trifluoromethylphenoxy)phenol in 3 ml of N,N-dimethylformamide with stirring and ice-cooling. After 30 minutes, a solution of 0.28 g of 2-chloro-5-chloromethylpyridine in 5 ml of N,N-dimethylformamide was added thereto at room temperature, followed by stirring at the same temperature for 10 hours. The reaction mixture was poured into ice-water and extracted twice with 100 ml of ethyl acetate. The extracts were combined together, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. The residue was subjected to silica gel chromatography to give 2-chloro-5-[2-chloro-4-(3-trifluoromethylphenoxy)phenoxy]methylpyridine.

Production Example 3 (Production of Compound No. 34)

To a solution of 0.07 g of sodium hydride (60 % oil dispersion) in 10 ml of N,N-dimethylformamide, there is added dropwise a solution of 0.50 g of 2-chloro-4-(3-bromo-phenoxy)phenol in 3 ml of N,N-dimethylformamide with stirring and ice-cooling. After 30 minutes, a solution of 0.27 g of 2-chloro-5-chloromethylpyridine in 5 ml of N,N-dimethylformamide is added thereto at room temperature, followed by stirring at the same temperature for 10 hours. The reaction mixture is poured into ice-water and extracted twice with 100 ml of ethyl acetate. The extracts are combined together, washed with water, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. The residue is subjected to silica gel chromatography to give 2-chloro-5-[4-(3-bromophenoxy)-2-chlorophenoxy]methylpyridine.

Production Example 4 (Production of Compound No. 17)

A mixture of 0.82 g of 2-chloro-5-phenoxyphenol, 0.54 g of 2-methyl-5-hydroxymethylpyridine, 1.01 g of triphenylphosphine, 0.65 g of diethyl azadicarboxylate and 90 ml of tetrahydrofuran was stirred at room temperature. After 48 hours, the reaction mixture was concentrated, and 50 ml of diethyl ether were added thereto. The precipitate was removed by filtration, and the filtrate was concentrated. The residue was subjected to silica gel chromatography to give 0.66 g of 2-methyl-5-(2-chloro-5-phenoxyphenoxy)-methylpyridine. Yield, 51 %. m.p., 64.9 °C

Production Example 5 (Production of Compound No. 35)

A mixture of 0.80 g of 2-chloro-5-phenoxyphenol, 0.57 g of 2-chloro-5-(1-hydroxyethyl)pyridine, 0.95 g of triphenylphosphine, 0.63 g of diethyl azadicarboxylate and 100 ml of tetrahydrofuran is stirred at room temperature. After 48 hours, the reaction mixture is concentrated, and 50 ml of diethyl ether are added thereto. The precipitate is removed by filtration, and the filtrate is concentrated. The residue is subjected to

silica gel chromatography to give 2-chloro-5-[1-(2-chloro-4-phenoxyphenoxy)ethyl]pyridine.

Some examples of the pyridine derivatives (I) as produced in the same manner as above are shown in Table 2.

5

Table 2

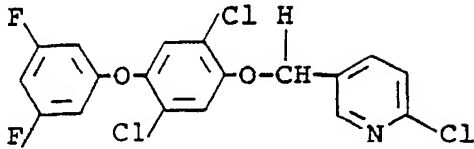
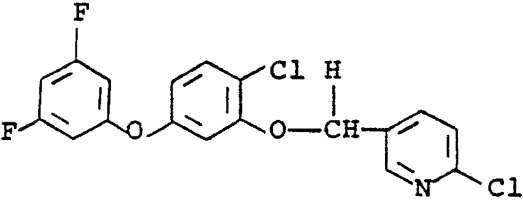
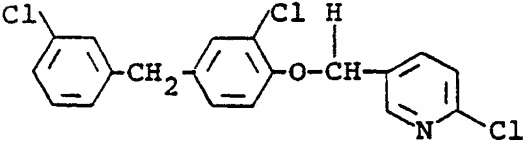
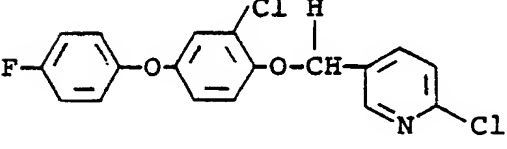
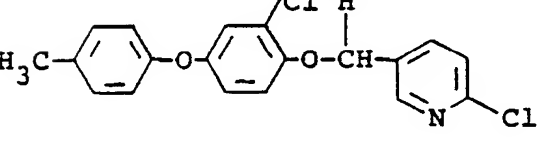
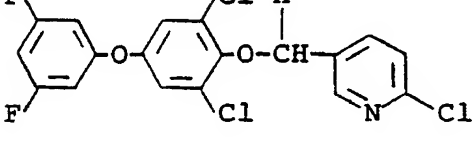
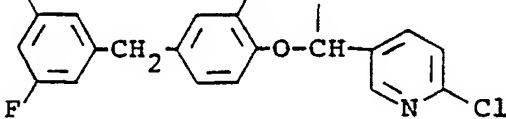
Compound No.	Chemical structure	Physical constant
1		n_D^{22} 1.6173
2		n_D^{25} 1.5843
3		m.p. 109.4°C
4		m.p. 65.6°C
5		m.p. 83.5°C
6		m.p. 93.9°C

50

55

Compound No.	Chemical structure	Physical constant
7		n_D^{22} 1.6142
8		m.p. 98.4°C
9		m.p. 79.3°C
10		n_D^{22} 1.6090
11		m.p. 77.2°C
12		m.p. 49.6°C
13		n_D^{23} 1.6083

Compound No.	Chemical structure	Physical constant
14		n_D^{23} 1.5939
15		n_D^{22} 1.6157
16		m.p. 88.1°C
17		m.p. 64.9°C
18		m.p. 89.8°C
19		n_D^{24} 1.6208
20		m.p. 93.7°C

Compound No.	Chemical structure	Physical constant
21		m.p. 84.2°C
22		m.p. 95.5°C
23		m.p. 79.9°C
24		m.p. 73.6°C
25		m.p. 75.4°C
26		m.p. 79.3°C
27		m.p. 110.6°C

Compound No.	Chemical structure	Physical constant
28		m.p. 94.2°C
29		m.p. 108.1°C
30		n_D^{22} 1.6201
31		m.p. 103.8°C
32		m.p. 76.1°C
33		
34		

Compound No.	Chemical structure	Physical constant
35		
36		m.p. 105.2°C
37		n_D^{23} 1.6352
38		m.p. 98.5°C
39		m.p. 50.8°C
40		m.p. 69.5°C

50 Production Example 6 (Production of Compound No. 42 as an intermediate)

To a solution of 5.0 g p-(3,5-difluorophenoxy)phenol in 50 mg of carbon tetrachloride, there were added dropwise 2.45 g of tert-butylhypochlorous acid with stirring and ice-cooling, and stirring was continued at room temperature for 5 hours. The reaction mixture was concentrated and extracted with 200 ml of ethyl acetate. The extract was washed with 5 % aqueous solution of sodium bicarbonate, dried over anhydrous magnesium sulfate and concentrated under reduced pressure. The residue was subjected to silica gel chromatography to give 5.31 g of 2-chloro-4-(3,5-difluorophenoxy)phenol. Yield, 92 %. n_D^{25} : 1.5639

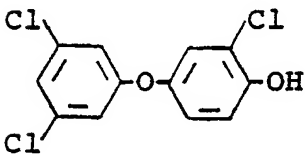
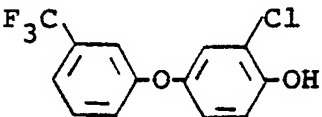
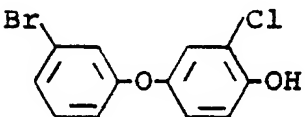
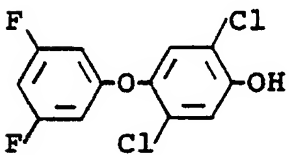
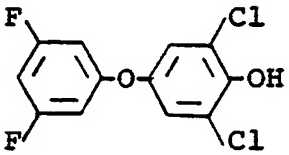
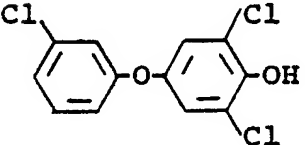
The compounds useful as the intermediates, which are produced in the same manner as above, are

shown in Table 3.

Table 3

Compound No.	Chemical structure	Physical constant
41		n_D^{25} 1.5846
42		n_D^{25} 1.5639
43		n_D^{25} 1.5960

Compound No.	Chemical structure	Physical constant
5 44		n_D^{25} 1.5960
10 45		n_D^{25} 1.5886
15 46		n_D^{25} 1.5934
20 47		$n_D^{21.5}$ 1.6098
30 48		$n_D^{21.4}$ 1.5670
35 49		n_D^{23} 1.5170
40 50		n_D^{25} 1.5930
45 51		n_D^{23} 1.6109

Compound No.	Chemical structure	Physical constant
52		n_D^{24} 1.5824
53		
54		
55		n_D^{23} 1.5650
56		m.p. 76.8°C
57		m.p. 80.6°C

In Formulation Examples as set forth below, parts and % are all by weight. The compound numbers
correspond to those as shown in Table 2.

Formulation example 1 (Emulsifiable concentrate)

To a solution of 10 parts of each of Compounds Nos. 1 to 40 in 35 parts of xylene and 35 parts of
dimethylformamide, 14 parts of polyoxyethylene styrylphenyl ether and 6 parts of calcium dodecylben-
zenesulfonate are added, and the resultant mixture is thoroughly mixed while stirring to give an emulsifiable
concentrate containing the active ingredient in 10 %.

Formulation Example 2 (Wettable powder)

Twenty parts of each of Compounds Nos. 1 to 40 are added to a mixture of 4 parts of sodium
laurylsulfate, 2 parts of calcium ligninsulfonate, 20 parts of fine powders of synthetic hydrated silica and 54
parts of diatomaceous earth, and the resultant mixture is stirred in a mixer to give a wettable powder

containing the active ingredient in 20 %.

Formulation Example 3 (Granules)

5 Five parts of sodium dodecylbenzenesulfonate, 30 parts of bentonite and 60 parts of clay are added to 5 parts of each of Compound Nos. 1, 2, 7, 10, 13 to 15, 19, 30 and 37, and the resultant mixture is pulverized and kneaded with a suitable amount of water. The mixture is granulated in a granulator and air-dried to give granules containing the active ingredient in 5 %.

10 Formulation Example 4 (Granules)

Five parts of fine powders of synthetic hydrated silica, 5 parts of sodium dodecylbenzenesulfonate, 30 parts of bentonite and 55 parts of clay are added to 5 parts of each of Compound Nos. 3 to 6, 8, 9, 11, 12, 16 to 18, 20 to 29, 31, 32, 36, 38 to 40, and the resultant mixture is pulverized and kneaded with a suitable
15 amount of water. The mixture is granulated in a granulator and air-dried to give granules containing the active ingredient in 5 %.

Formulation Example 5 (Dusts)

20 To a mixture of 1 part of fine powders of synthetic hydrated silica, 1 part of an aggregating agent ("Driless B" manufactured by Sankyo Co., Ltd.) and 7.7 parts of clay, 0.3 part of each of Compound Nos. 1, 2, 7, 10, 13 to 15, 19, 30 and 37 is added, and the resultant mixture is well pestled in a mortar and further stirred in a mixer. To the thus obtained mixture, there are added 90 parts of cut clay, followed by mixing to give dusts containing the active ingredient in 0.3 %.

25 Formulation Example 6 (Dusts)

A mixture of 0.3 part of each of Compound Nos. 3 to 6, 8, 9, 11, 12, 16 to 18, 20 to 29, 31, 32, 36 and 38 to 40 and 0.03 part of fine powders of synthetic hydrated silica is stirred well in a mixer and pulverized
30 by the aid of a centrifugal pulverizer. To the resultant mixture, 0.97 part of fine powders of synthetic hydrated silica, 1 part of "Driless B" and 7.7 parts of clay are added, and the resulting mixture is pestled in a mortar and stirred in a mixer. Ninety parts of cut clay are added thereto, and further mixing is effected in a sack to give dusts containing the active ingredient in 0.3 %.

35 Formulation Example 7 (Dusts)

A mixture of 0.3 part of each of Compound Nos. 1, 2, 7, 10, 13 to 15, 19, 30 and 37, 2 parts of BPMC (O-sec-buthylphenyl N-methylcarbamate) as a carbamate insecticide, 3 parts of fine powders of synthetic hydrated silica, 1 part of "Driless B" and 3.7 parts of clay are pestled in a mortar and stirred in a mixer.
40 Then, 90 parts of cut clay are added thereto, and the resultant mixture is further mixed in a sack to give dusts.

Formulation Example 8 (Dusts)

45 A mixture of 0.3 part of each of Compound Nos. 3 to 6, 8, 9, 11, 12, 16 to 18, 20 to 29, 31, 32, 36 and 38 to 40 and 0.03 part of fine powders of synthetic hydrated silica is stirred in a mixer and pulverized by a centrifugal pulverizer. After addition of 2 parts of BPMC (O-sec-butylphenyl N-methylcarbamate), 2.97 parts of fine powders of synthetic hydrated silica, 1 part of "Driless B" and 3.7 parts of clay thereto, the resultant mixture is pestled in a mortar and stirred in a mixer. Then, 90 parts of cut clay are added thereto, and the
50 resultant mixture is further mixed in a sack to give dusts.

Formulation Example 9 (Dusts)

To a solution of 1 part of each of Compound Nos. 1 to 40 in an appropriate amount of acetone, 5 parts
55 of fine powders of synthetic hydrated silica, 0.3 part of PAP (acidic isopropyl phosphate) and 93.7 parts of clay are added, and the resultant mixture is stirred in a mixer, followed by evaporation of acetone to give dusts containing the active ingredient in 1 %.

Formulation Example 10 (Dusts)

To 40 parts of an aqueous solution containing 2 parts of polyvinyl alcohol, 10 parts of each of Compound Nos. 1, 2, 7, 10, 13 to 15, 19, 30 and 37 are added, and the resultant mixture is stirred in a mixer. To the thus obtained dispersion, 40 parts of an aqueous solution containing 0.05 part of xanthane gum and 0.1 part of aluminum magnesium silicate are added, followed by addition of 10 parts of propylene glycol. The mixture is gently stirred to give a flowable concentrate containing the active ingredient in 10 %.

Formulation Example 11 (Flowable concentrate)

To 28.5 parts of an aqueous solution containing 2 parts of polyvinyl alcohol, 20 parts of each of Compound Nos. 3 to 6, 8, 9, 11, 12, 16 to 18, 20 to 29, 31, 32, 36 and 38 to 40 and 1.5 parts of sorbitan trioleate are added, and the resultant mixture is finely pulverized by the aid of a sand grinder to give particles of less than 3 microns in average particle size. To the resultant mixture, 40 parts of an aqueous solution containing 0.05 part of xanthane gum and 0.1 part of aluminum magnesium silicate are added, followed by addition of 10 parts of propylene glycol. The mixture is gently stirred to give a flowable concentrate containing the active ingredient in 20 %.

Formulation Example 12 (Oil spray)

Into a mixture of 5 parts of xylene and 5 parts of trichloroethane, 0.1 part of each of Compound Nos. 1 to 40 is dissolved, and the resultant solution is mixed with 89.9 parts of deodorized kerosene to give an oil spray containing the active ingredient in 0.1 %.

Formulation Example 13 (Oil-based aerosol)

A solution of 0.1 part of each of Compound Nos. 1 to 40, 0.2 part of tetramethrin (2,2-dimethyl-3-(2-methyl-1-propenyl)cyclopropanecarboxylic acid (1,2,3,4,5,7-hexahydro-1,3-dioxo-2H-isoindol-2-yl)methyl ester) and 0.1 part of d-phenothrin (2,2-dimethyl-3-(2-methyl-1-propenyl)cyclopropanecarboxylic acid (3-phenoxyphenyl)methyl ester) in a mixture of 10 parts of trichloroethane and 59.6 parts of deodorized kerosene is filled in an aerosol container. After provision of a valve, 30 parts of a propellant (liquefied petroleum gas) is filled through the valve under reduced pressure to give an oil-based aerosol.

Formulation Example 14 (Water-based aerosol)

A solution of 0.2 part of each of Compound Nos. 1 to 40, 0.2 part of d-allethrin, 0.2 part of d-phenothrin, 5 parts of xylene, 3.4 parts of deodorized kerosene and 1 part of an emulsifier ("ATMOS 300"®, Atlas Chemical Co., Ltd.) in 50 parts of distilled water is filled in an aerosol container. After provision of a valve, 40 parts of a propellant (liquefied petroleum gas) is filled through the valve under reduced pressure to give a water-based aerosol.

Formulation Example 15 (Fumigant)

Each of Compound Nos. 1 to 40 (100 mg) is dissolved in an appropriate amount of acetone, and the resultant solution is impregnated with a porous ceramic plate (4.0 x 4.0 x 1.2 cm) to give a fumigant.

The following Test Examples show some of test results which support the controlling effect of the pyridine derivatives (I) on insect pests. The compound numbers correspond to those in Table 2. The compound used for comparison is as follows:

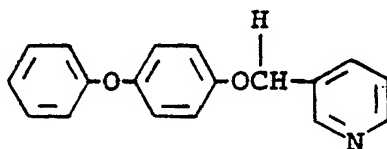
50

Compound
symbol · Chemical structure

Remarks

55

A



Compound disclosed
in JP-A-50-18628
(laid open)

Test Example 1

Metamorphosis inhibitory activity against brown rice planthopper nymphae:-

An emulsifiable concentrate prepared according to Formulation Example 1 was diluted with water to make a predetermined concentration. The dilution was sprayed onto rice plants cultivated in polyethylene cups at a rate of 20 ml/2 pots on a turning table. After air-drying, the plants were infested with about ten 3rd instar nymphae of brown rice planthopper (*Nilaparvata lugens*). After 10 days, the number of normal adults was counted to obtain an emergence inhibitory rate. The results are shown in Table 4.

10

Table 4

	<u>Compound No.</u>	<u>Concentration (ppm)</u>	<u>Inhibitory rate (%)</u>
15	1	50	100
		5	100
	2	50	100
	3	50	100
	4	50	100
20	5	50	100
		5	100
		0.5	100
	6	50	100
		5	100
25		0.5	100
		0.05	100
	7	50	100
		5	100
		0.5	100
30	8	50	100
		5	100
		0.5	100
	9	50	100
		5	100
35		0.5	100
		0.05	100
	10	50	100
		5	100
40	11	50	100
		5	100
		0.5	100
		0.05	100
	12	50	100
		5	100
45	13	50	100
		5	100
		0.5	100
	14	50	100
		5	100
50		0.5	100
		0.05	100
	15	50	100
	16	50	100
55		5	100

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	<u>Compound No.</u>	<u>Concentration (ppm)</u>	<u>Inhibitory rate (%)</u>
5	17	50	100
		5	100
		0.5	100
	19	50	100
		5	100
		0.5	100
10	20	0.05	100
		50	100
		5	100
	21	0.5	100
		0.05	100
		50	100
15	22	5	100
		0.5	100
		0.05	100
	23	50	100
		5	100
		0.5	100
20	24	0.05	100
		50	100
		5	100
	25	0.5	100
		0.05	100
		50	100
25	26	5	100
		50	100
		50	100
	27	5	100
		50	100
		5	100
30	28	0.5	100
		0.05	100
		50	100
	29	5	100
		50	100
		5	100
35	30	50	100
		50	100
		50	100
	31	5	100
		50	100
		5	100
40	32	50	100
		5	100
		5	100
	35	50	100
		5	100
		5	100
45	36	50	100
		5	100
		5	100
	37	50	100
		50	100
		50	100
50	38	5	100
		0.5	100
		0.05	100

55

	<u>Compound No.</u>	<u>Concentration (ppm)</u>	<u>Inhibitory rate (%)</u>
5	39	50	100
		5	100
		0.5	100
		0.05	100
	40	50	100
		5	100
10		0.5	100
	A	50	10

15 Test Example 2

Reproduction inhibitory activity against green rice leafhopper:-

An emulsifiable concentrate prepared according to Formulation Example 1 was diluted with water to make a predetermined concentration. The dilution was sprayed onto rice plants (about 20 cm in height) cultivated in plastic pots (1/5000 are in width) at a rate of 40 ml/2 pots on a turning table. After air-drying, the pots were covered with wire cages, and each 10 female and male adults of green rice leafhopper (*Nephotettix cincticeps*) were released in each of the cages. After 3 weeks, the number of nymphae was counted to obtain a reproduction inhibitory rate. The results are shown in Table 5.

25 Table 5

	<u>Compound No.</u>	<u>Concentration (ppm)</u>	<u>Inhibitory rate (%)</u>
30	6	100	100
	11	100	100
	20	100	100
	21	100	100
	22	100	100
35	27	100	99
	A	100	0

40 Test Example 3

Reproduction inhibitory activity against brown rice planthopper:-

An emulsifiable concentrate prepared according to Formulation Example 1 was diluted with water to make a predetermined concentration. The dilution was sprayed onto rice plants (about 20 cm in height) cultivated in plastic pots (1/5000 are in width) at a rate of 40 ml/2 pots on a turning table. After air-drying, the pots were covered with wire cages, and each 5 female and male adults of brown rice planthopper (*Nilaparvata lugens*) were released in each of the cages. After about 3 weeks, the number of nymphae was counted to obtain a reproduction inhibitory rate. The result are shown in Table 6.

50

55

Table 6

	<u>Compound No.</u>	<u>Concentration (ppm)</u>	<u>Inhibitory rate (%)</u>
5	6	100	100
	11	100	100
	13	100	100
10	20	100	100
	21	100	100
	22	100	100
	27	100	100
15	A	100	0

Test Example 4

20 Reproduction inhibitory activity against cotton aphids:

An emulsifiable concentrate prepared according to Formulation Example 1 was diluted with water to make a predetermined concentration. The dilution was sprayed onto potted cotton plants (in a stage of 8-9 days after sowing) infested with 1st instar nymphae of cotton aphids (*Aphis gossypii*) at a rate of 30 ml/2 pots on a turning table. Before spraying and one week after spraying, the number of nymphae and adults
 25 was counted, and a reproduction inhibitory index was expressed by the following equation:

$$\begin{array}{lcl}
 \text{Reproduction} & & \text{Number of individuals one week} \\
 \text{inhibitory} & = & \text{after spraying per 2 pots} \\
 \text{index} & & \hline
 & & \text{Number of individuals before} \\
 & & \text{spraying per 2 pots}
 \end{array}$$

wherein the judgement of activity is based on the following standard:

- 35 A: less than 1 (excellent effect)
 B: from 1 to 3 (slight effect)
 C: more than 3 (little effect)
 D: same as in the untreated pots (no effect)

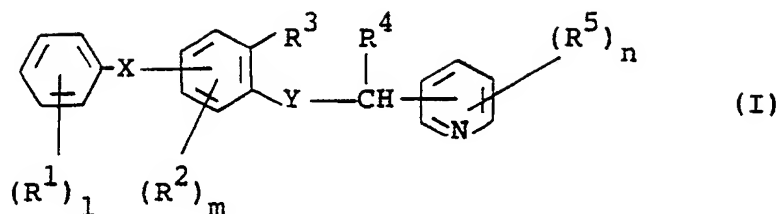
The results are shown in Table 7.

Table 7

	<u>Compound No.</u>	<u>Concentration (ppm)</u>	<u>Inhibitory rate (%)</u>
45	6	10	A
	11	10	A
	A	10	D

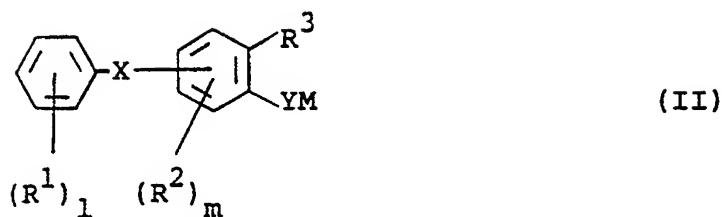
Claims

1. A pyridine derivative of the formula (I)

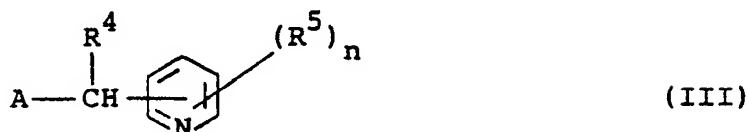


wherein R¹ is, the same or different, each a hydrogen atom, a halogen atom, a C₁-C₄ alkyl group, a C₁-C₄ haloalkyl group, a C₁-C₃ alkoxy group, a C₁-C₃ haloalkoxy group, a cyano group or a nitro group; R² is, the same or different, each a hydrogen atom, a halogen atom or a methyl group; R³ is a halogen atom or a C₁-C₃ alkyl group; R₄ is a hydrogen atom or a C₁-C₃ alkyl group; R⁵ is, the same or different, each a halogen atom, a C₁-C₃ alkyl group, a C₁-C₃ haloalkyl group, a C₁-C₃ alkoxy group or a C₁-C₃ haloalkoxy group; X is an oxygen atom, a sulfur atom, a carbonyl group, a sulfinyl group, a sulfonyl group, an imino group or a methylene group; Y is an oxygen atom or a sulfur atom; l is an integer of 1 to 5; m is an integer of 1 to 3; and n is an integer of 1 to 4.

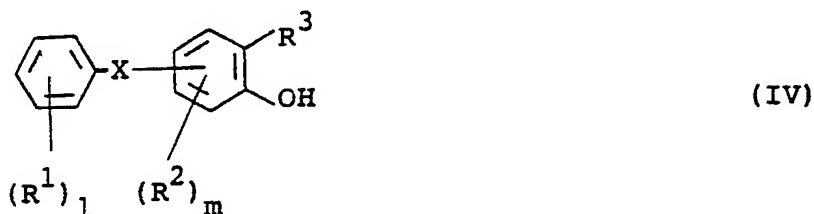
2. The pyridine derivative according to claim 1, wherein R¹ is, the same or different, each a hydrogen atom, a fluorine atom, a chlorine atom, a bromine atom, a methyl group or a trifluoromethyl group.
3. The pyridine derivative according to claim 1, wherein R¹ is, the same or different, each a hydrogen atom, a fluorine atom or a chlorine atom, R² is a hydrogen atom or a chlorine atom, R³ is a fluorine atom, a chlorine atom, a bromine atom or a methyl group, X is an oxygen atom or a methylene group, Y is an oxygen atom, l is an integer of 1 or 2, m is an integer of 1 and n is an integer of 1 or 2.
4. The pyridine derivative according to claim 1, wherein (R¹)₁ is 2,4-F₂, 3,5-F₂, 3,4-Cl₂, 3-F or 3-Cl, R² is a hydrogen atom, R³ is a chlorine atom, R₄ is a hydrogen atom, R⁵ is a chlorine atom or a methyl group, X is an oxygen atom or a methylene group, Y is an oxygen atom and m and n are each an integer of 1.
5. The pyridine derivative according to claim 1, wherein R¹ is a hydrogen atom, R² is a hydrogen atom, R₃ is a chlorine atom, R₄ is a hydrogen atom, R⁵ is a chlorine atom or a methyl group, X is a methylene group, Y is an oxygen atom and m and n are each an integer of 1.
6. The pyridine derivative according to claim 1, wherein (R¹)₁ is 2,4-F₂, 3,5-F₂ or 3-Cl, R² is a hydrogen atom, R³ is a chlorine atom, R₄ is a hydrogen atom, R⁵ is a chlorine atom or a methyl group, X is an oxygen atom or a methylene group, Y is an oxygen atom and m and n are each an integer of 1.
7. The pyridine derivative according to claim 1, wherein (R¹)₁ is 3-F, R² is a hydrogen atom, R³ is a chlorine atom, R₄ is a hydrogen atom, R⁵ is a chlorine atom or a methyl group, X is a methylene group, Y is an oxygen atom and m and n are each an integer of 1.
8. The pyridine derivative according to claim 1, which is 2-chloro-5-[2-chloro-4-(3,5-difluorophenoxy)-phenoxy]methylpyridine,
 2-methyl-5-[2-chloro-4-(3,5-difluorophenoxy)phenoxy]methylpyridine,
 2-methyl-5-[2-chloro-4-(3,4-dichlorophenoxy)phenoxy]methylpyridine,
 2-chloro-5-[2-chloro-4-(3-fluorobenzyl)phenoxy]methylpyridine,
 2-chloro-5-[2,5-dichloro-4-(3,5-difluorophenoxy)phenoxy]methylpyridine,
 2-chloro-5-[2-chloro-5-(3,5-difluorophenoxy)phenoxy]methylpyridine,
 2-chloro-5-[2-chloro-4-(3,5-difluorobenzyl)phenoxy]methylpyridine,
 2-methyl-5-[2-chloro-4-(2,4-difluorophenoxy)phenoxy]methylpyridine, or
 2-chloro-5-[2-chloro-4-(3-chlorophenoxy)phenoxy]methylpyridine.
9. A process for producing a pyridine derivative of the formula (I) as defined in claim 1, which comprises
 (a) reacting a phenol compound of the formula (II)



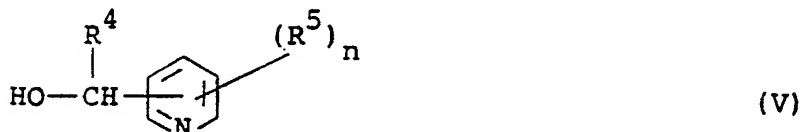
10 wherein M is an alkali metal atom or a hydrogen atom and R¹, R², R³, X, Y, l and m are each as defined in claim 1 with a halide of the formula (III)



20 wherein A is a halogen atom and R⁴, R⁵ and n are each as defined in claim 1 to give the pyridine derivative (I); or
(b) reacting a phenol compound of the formula (IV)



35 wherein R¹, R², R³, X, l and m are each as defined in claim 1 with an alcohol of the formula (V)



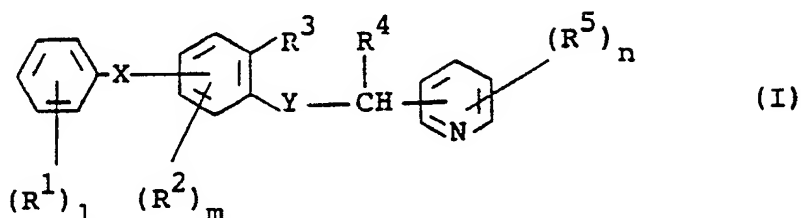
45 wherein R⁴, R⁵ and n are each as defined in claim 1 to give the pyridine derivative (I) wherein Y is an oxygen atom.

10. The process according to claim 9, wherein the reaction in the step (a) is carried out in an inert solvent in the presence of a base at a temperature of from about -20 °C to the boiling point of the inert solvent.
- 50 11. The process according to claim 9, wherein the reaction in the step (b) is carried out in an inert solvent in the presence of a dehydrating catalyst or agent at a temperature of from about -20 to 200 °C or the boiling point of the inert solvent.
12. A composition for controlling insect pests which comprises the pyridine derivative of the formula (I) according to any of claims 1 to 8 and optionally an inert carrier.
- 55 13. A method for controlling insect pests which comprises applying the pyridine derivative of the formula (I) according to any of claims 1 to 8 to the insect pests or the locus where the insect pests propagate.

14. Use of the pyridine derivative of the formula (I) according to any of claims 1 to 8 as an agent for controlling insect pests.

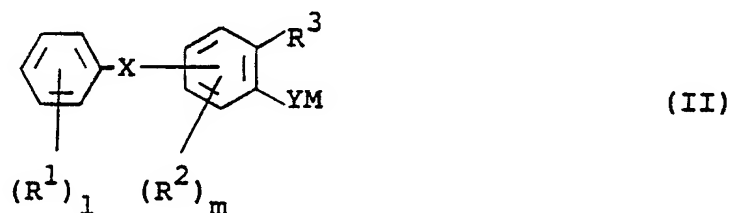
Claims for the following Contracting State: ES

1. A process for producing a pyridine derivative of the formula (I)

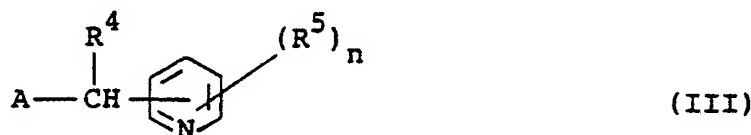


wherein R^1 is, the same or different, each a hydrogen atom, a halogen atom, a C_1 - C_4 alkyl group, a C_1 - C_4 haloalkyl group, a C_1 - C_3 alkoxy group, a C_1 - C_3 haloalkoxy group, a cyano group or a nitro group; R^2 is, the same or different, each a hydrogen atom, a halogen atom or a methyl group; R^3 is a halogen atom or a C_1 - C_3 alkyl group; R^4 is a hydrogen atom or a C_1 - C_3 alkyl group; R^5 is, the same or different, each a halogen atom, a C_1 - C_3 alkyl group, a C_1 - C_3 haloalkyl group, a C_1 - C_3 alkoxy group or a C_1 - C_3 haloalkoxy group; X is an oxygen atom, a sulfur atom, a carbonyl group, a sulfinyl group, a sulfonyl group, an imino group or a methylene group; Y is an oxygen atom or a sulfur atom; l is an integer of 1 to 5; m is an integer of 1 to 3; and n is an integer of 1 to 4, which comprises:

(a) reacting a phenol compound of the formula (II)

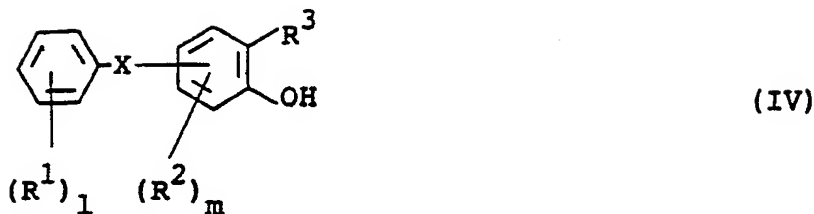


wherein M is an alkali metal atom or a hydrogen atom and R^1 , R^2 , R^3 , X, Y, l and m are each as defined above with a halide of the formula (III)

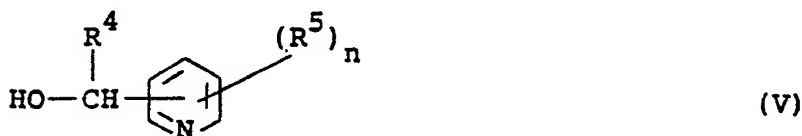


wherein A is a halogen atom and R^4 , R^5 and n are each as defined above to give the pyridine derivative (I); or

(b) reacting a phenol compound of the formula (IV)



10 wherein R¹, R², R³, X, l and m are each as defined above with an alcohol of the formula (V)



20 wherein R⁴, R⁵ and n are each as defined above to give the pyridine derivative (I) wherein Y is an oxygen atom.

2. The process according to claim 1,
 25 wherein R¹ is, the same or different, each a hydrogen atom, a fluorine atom, a chlorine atom, a bromine atom, a methyl group or a trifluoromethyl group.
3. The process according to claim 1,
 wherein R¹ is, the same or different, each a hydrogen atom, a fluorine atom or a chlorine atom, R² is a hydrogen atom or a chlorine atom, R³ is a fluorine atom, a chlorine atom, a bromine atom or a methyl group, X is an oxygen atom or a methylene group, Y is an oxygen atom, l is an integer of 1 or 2, m is an integer of 1 and n is an integer of 1 or 2.
4. The process according to claim 1,
 35 wherein (R¹)₁ is 2,4-F₂, 3,5-F₂, 3,4-Cl₂, 3-F or 3-Cl, R² is a hydrogen atom, R₃ is a chlorine atom, R⁴ is a hydrogen atom, R⁵ is a chlorine atom or a methyl group, X is an oxygen atom or a methylene group, Y is an oxygen atom and m and n are each an integer of 1.
5. The process according to claim 1,
 40 wherein R¹ is a hydrogen atom, R² is a hydrogen atom, R₃ is a chlorine atom, R⁴ is a hydrogen atom, R⁵ is a chlorine atom or a methyl group, X is a methylene group, Y is an oxygen atom and m and n are each an integer of 1.
6. The process according to claim 1,
 45 wherein (R¹)₁ is 2,4-F₂, 3,5-F₂ or 3-Cl, R² is a hydrogen atom, R₃ is a chlorine atom, R⁴ is a hydrogen atom, R⁵ is a chlorine atom or a methyl group, X is an oxygen atom or a methylene group, Y is an oxygen atom and m and n are each an integer of 1.
7. The process according to claim 1,
 50 wherein (R¹)₁ is 3-F, R² is a hydrogen atom, R₃ is a chlorine atom, R⁴ is a hydrogen atom, R⁵ is a chlorine atom or a methyl group, X is a methylene group, Y is an oxygen atom and m and n are each an integer of 1.
8. The process according to claim 1, wherein the pyridine derivative is
 55 2-chloro-5-[2-chloro-4-(3,5-difluorophenoxy)phenoxy]methylpyridine,
 2-methyl-5-[2-chloro-4-(3,5-difluorophenoxy)phenoxy]methylpyridine,
 2-methyl-5-[2-chloro-4-(3,4-dichlorophenoxy)phenoxy]methylpyridine,
 2-chloro-5-[2-chloro-4-(3-fluorobenzyl)phenoxy]methylpyridine,
 2-chloro-5-(2,5-dichloro-4-(3,5-difluorophenoxy)phenoxy]methylpyridine,

2-chloro-5-[2-chloro-5-(3,5-difluorophenoxy)phenoxy]methylpyridine,
 2-chloro-5-[2-chloro-4-(3,5-difluorobenzyl)phenoxy]methylpyridine,
 2-methyl-5-[2-chloro-4-(2,4-difluorophenoxy)phenoxy]methylpyridine, or
 2-chloro-5-[2-chloro-4-(3-chlorophenoxy)phenoxy]methylpyridine.

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9. The process according to any of claims 1 to 8, wherein the reaction in the step (a) is carried out in an inert solvent in the presence of a base at a temperature of from about -20 °C to the boiling point of the inert solvent.

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10. The process according to any of claims 1 to 8, wherein the reaction in the step (b) is carried out in an inert solvent in the presence of a dehydrating catalyst or agent at a temperature of from about -20 to 200 °C or the boiling point of the inert solvent.

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11. A composition for controlling insect pests which comprises the pyridine derivative of the formula (I) obtained by the process of any of claims 1 to 10 and optionally an inert carrier.

12. A method for controlling insect pests which comprises applying the pyridine derivative of the formula (I) obtained by the process of any of claims 1 to 10 to the insect pests or the locus where the insect pests propagate.

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13. Use of the pyridine derivative of the formula (I) obtained by the process of any of claims 1 to 10 as an agent for controlling insect pests.

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Application number

EP 91108483

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl. 5)
D, A	DE-A-2516331 (CIBA-GEIGY AG) * claims; page 4 *	1-14	C07D213/30 C07D213/32 A01N43/40
A	EP-A-0289919 (BASF AG) * claims; pages 10-25 *	1-14	
A	EP-A-0101288 (DOW CHEMICAL CO.) * abstract; table I *	1-8, 12-14	
A	EP-A-0299694 (SCHERING AGROCHEMICALS LTD.) * abstract; example 5 *	1-8, 12-14	
A	CHEMICAL ABSTRACTS vol. 112, no. 15, 9 April 1990, abstract no. 138739k, Columbus, Ohio, US; & JP - A - 01228929 (NIPPON SODA CO., LTD.) 12.09.1989	1-8, 12-14	
			TECHNICAL FIELDS SEARCHED (Int. Cl. 5)
			C07D213/00 C07D215/00 C07D233/00 C07D277/00
INCOMPLETE SEARCH			
<p>The Search Division considers that the present European patent application does not comply with the provisions of the European Patent Convention to such an extent that it is not possible to carry out a meaningful search into the state of the art on the basis of some of the claims.</p> <p>Claims searched completely: Claims searched incompletely: Claims not searched: Reason for the limitation of the search:</p> <p>As the drafting of the claims is not clear and concise (Art. 83-84 EPO) and encompasses such an enormous amount of products, a complete search is not possible on economic grounds (see Guidelines for Examination in the EPO, Part B, Chapter III, 2). So the search has been limited (Rule 45) to the examples.</p>			
Place of search Berlin		Date of completion of the search 24.07.1991	Examiner D. FRELON
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